

Mercury Source Tracing and Mechanistic Studies

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Unique Environmental

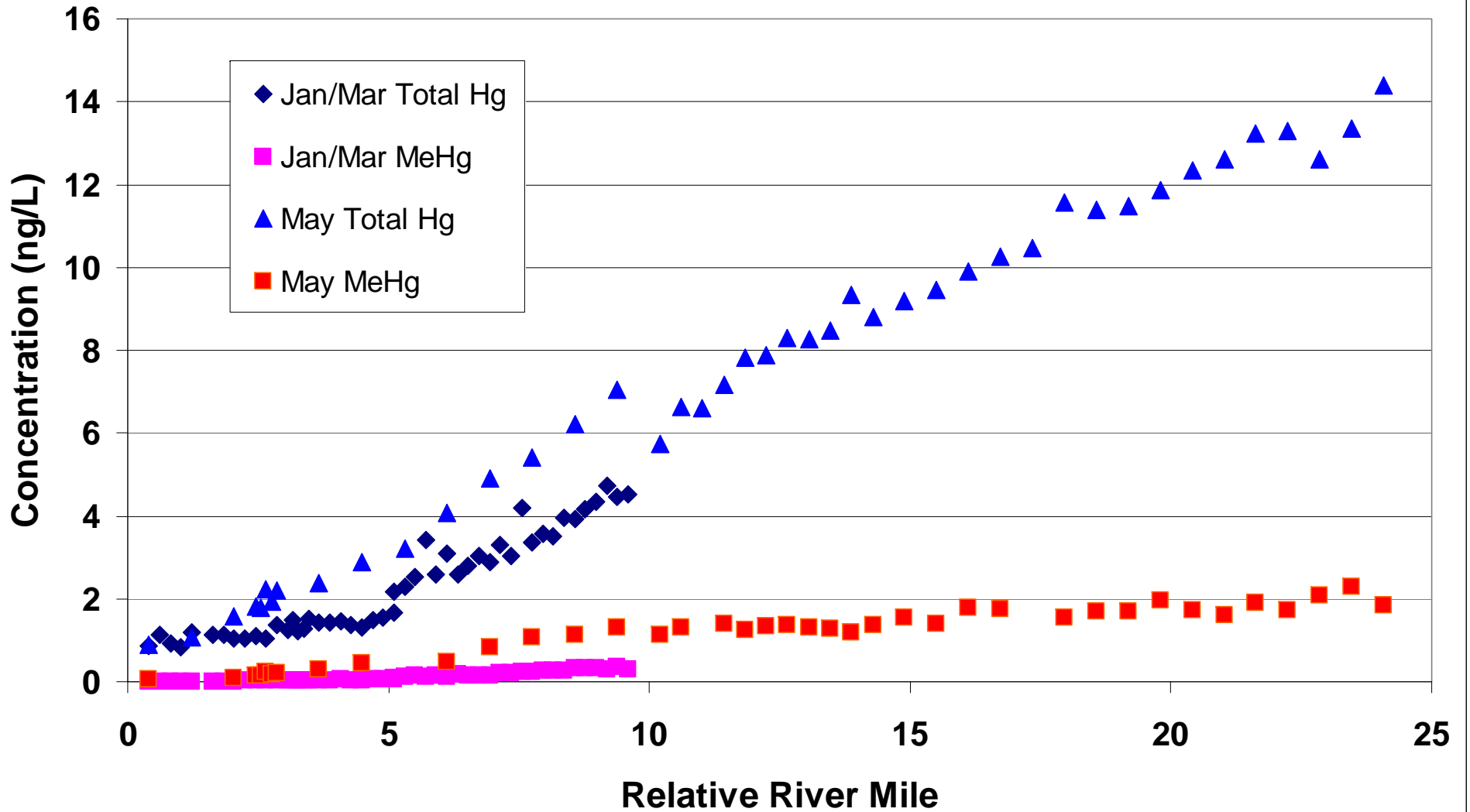
Synopsis

- Getting closer to answering the question “How is Hg getting into the South River in bioavailable form?”
 - Very likely not from point source(s)
 - Likely related to presence of Hg in floodplain/bank/bed solids in form(s) that can be released continuously into surface water
 - Role of shallow alluvial groundwater still being quantified

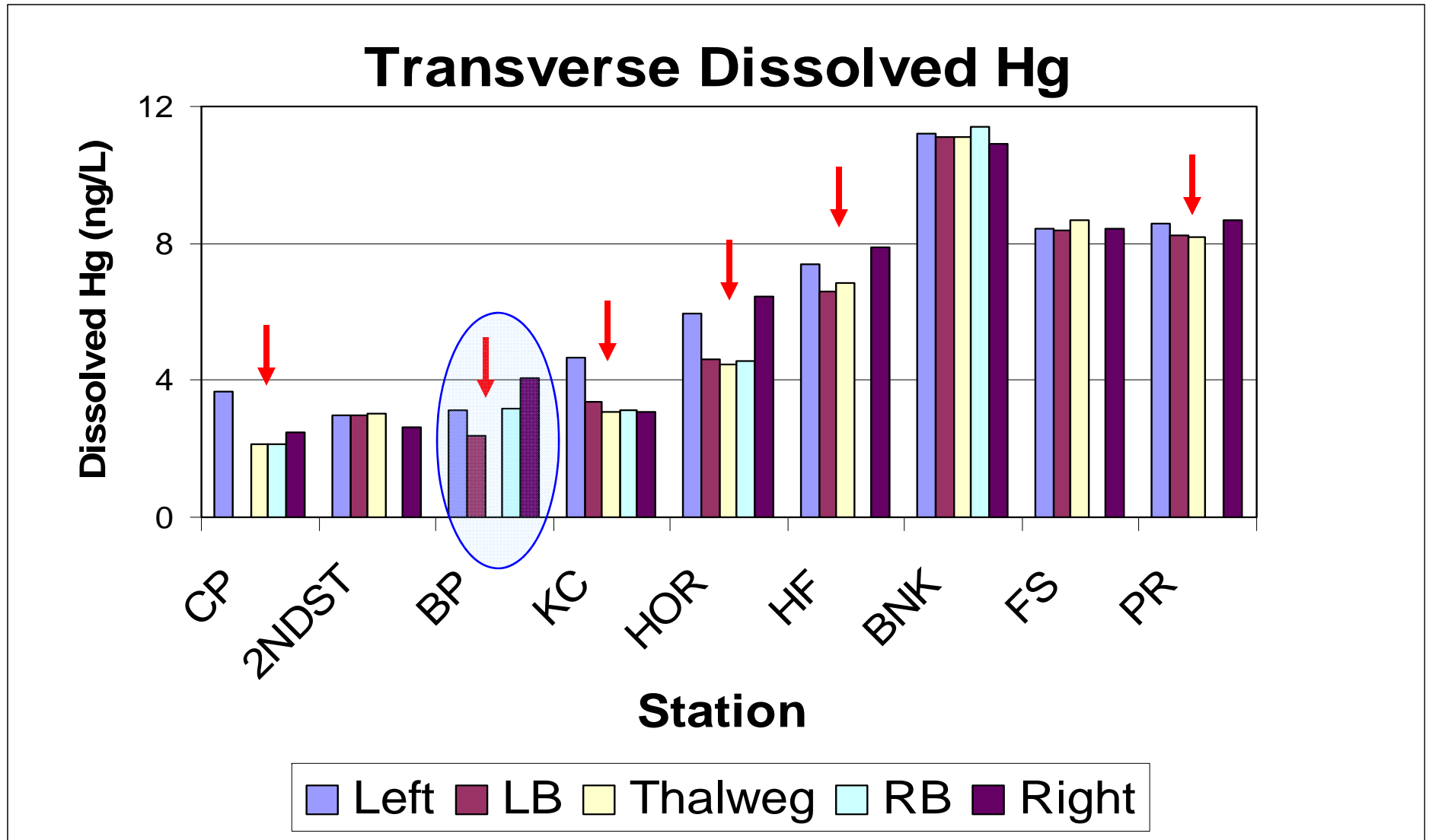
General Objectives

- Identify in-stream Hg source(s) within Plant reach (RRM-0.7 to 0.3).
 - Bank or bed source(s)
- Determine origin of higher dissolved Hg concentration near banks (@RRM2.2) relative to thalweg.
 - Desorption from bank/bed solids
 - Shallow groundwater
- Develop tools and procedures applicable elsewhere on the river.

2005 Close-Interval Surface Water Data (0.45 micron filtered)



Near-Bank Study Background



Special Clarification

- Elevated dissolved Hg results for near-bank SW samples **do not** rule out possible additional center stream inputs/fluxes.
- They **are** a strong indicator that a reach is an active input location for SW.
- They **are** a strong indicator that the near-bank locations are active input locations to SW.

Basic Park Study

Objectives: Confirm near-bank elevations of Hg and quantify/partition source between desorption from solids and alluvial groundwater.

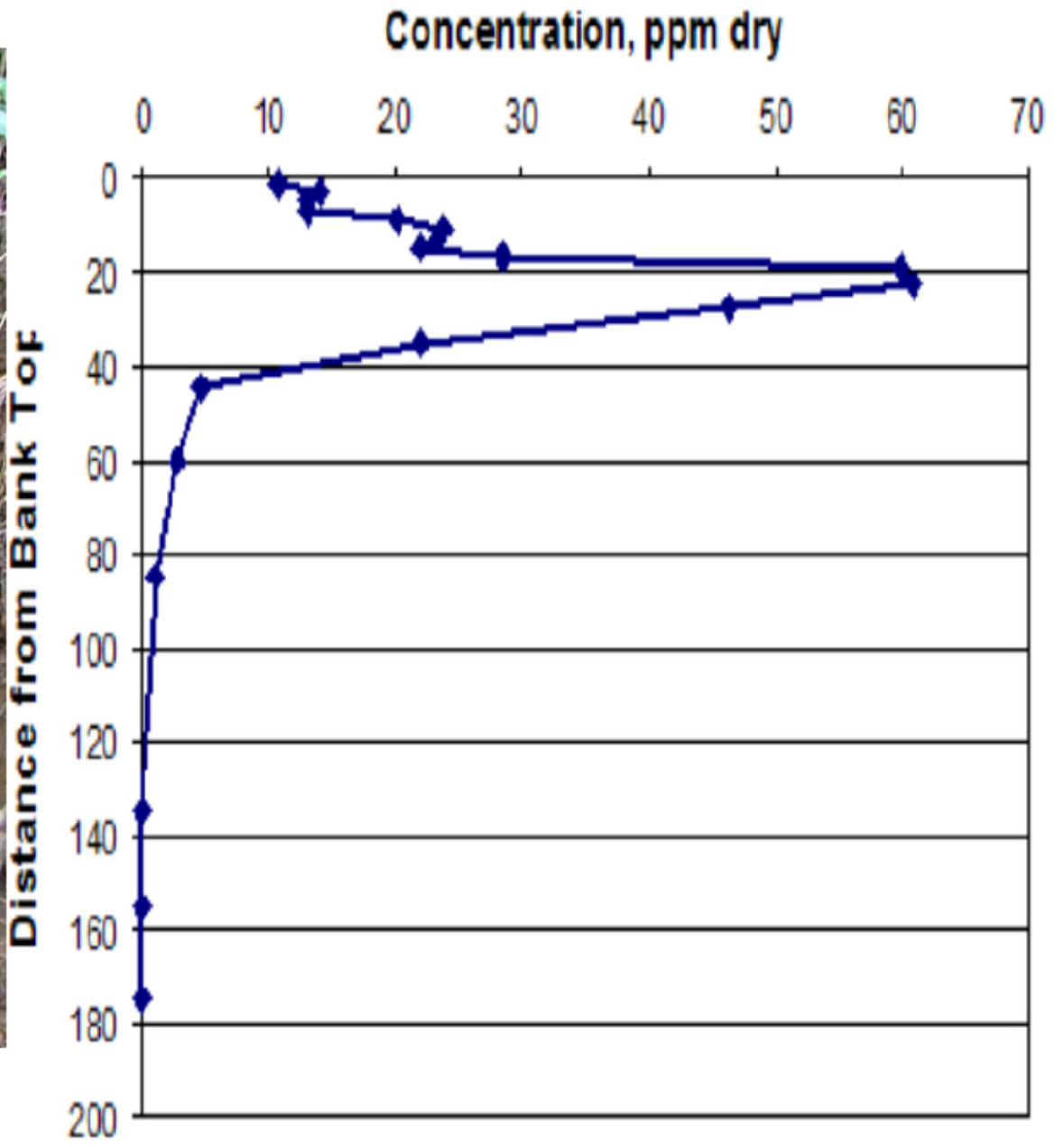
Near-Bank Study

Investigative Approaches in Phase I

- Chose location coincident with ecostudy
- Synoptic measurement of
 - Surface water Hg, transverse and longitudinal in study area, to verify elevated near-banks and provide baseline.
 - Near-bank sediment Hg
 - Pore water/groundwater Hg
 - “Exhaustive” extraction of near-bank sediments and nearby “parent” bank soils.







Sampled by University of Delaware-2005

Surface Water Pattern

Filtered THg (ng/L), May 2006

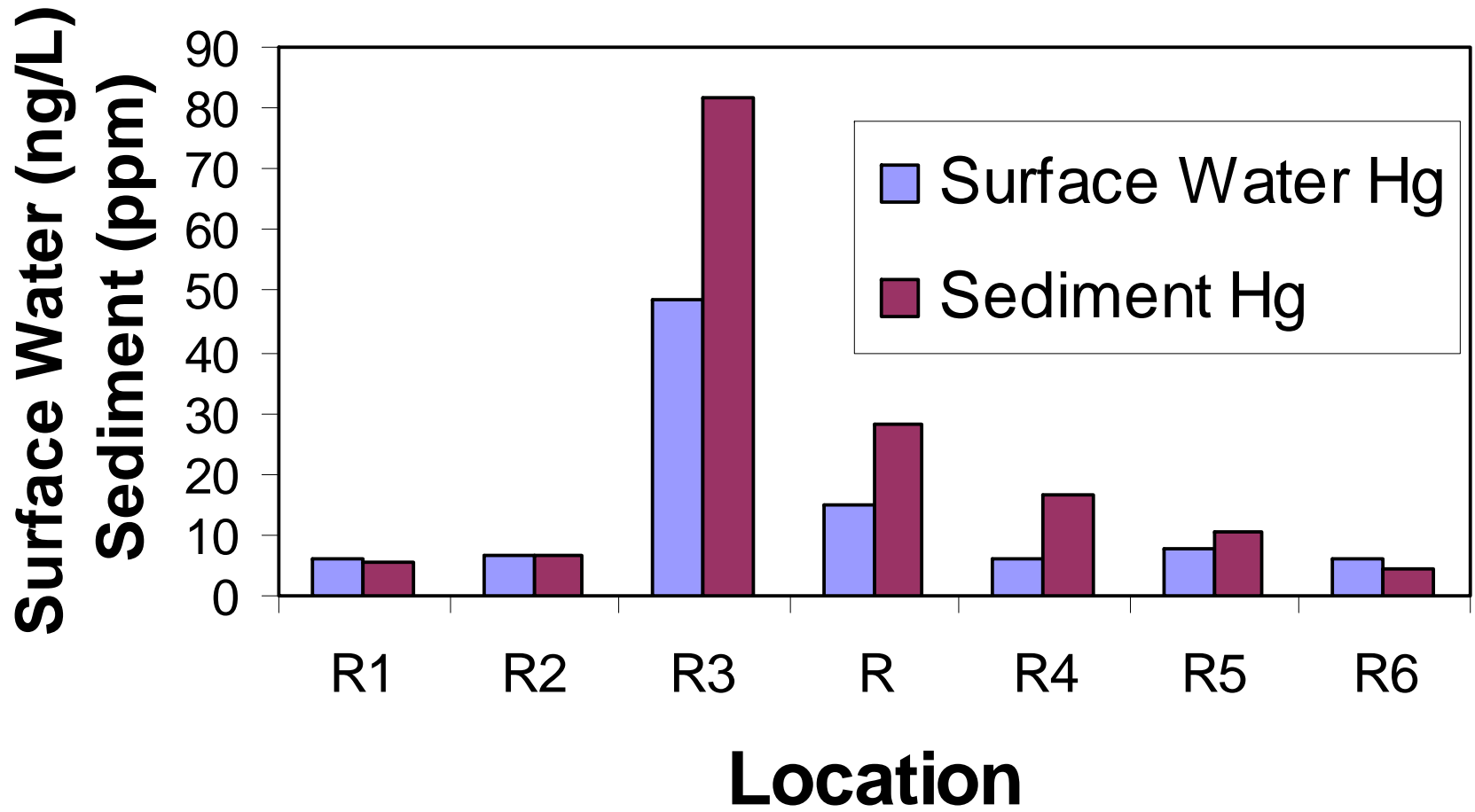


Shallow Sediment Pattern

THg ($\mu\text{g/g}$), **MHg (ng/g)**

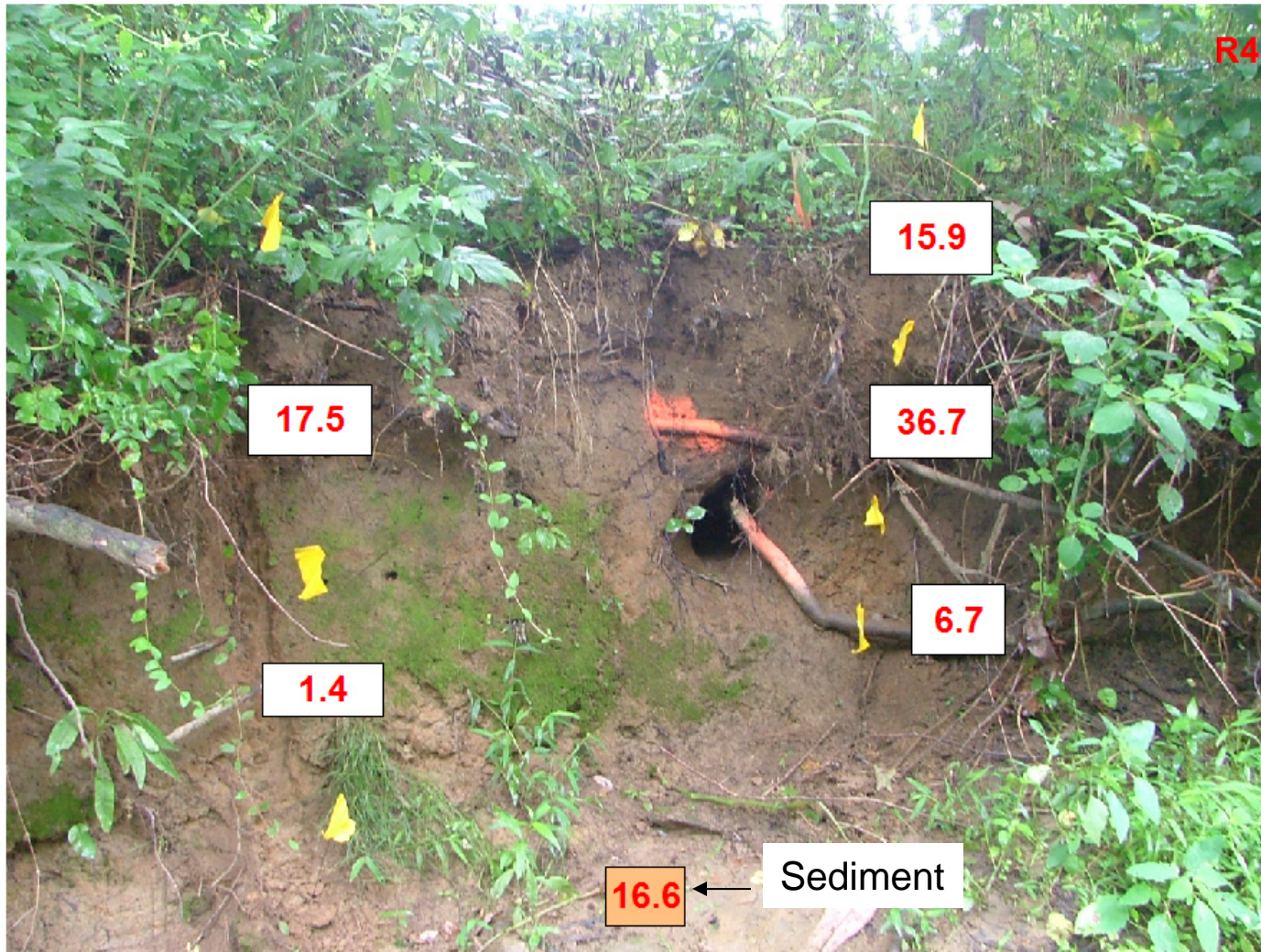


Basic Park - May 2006



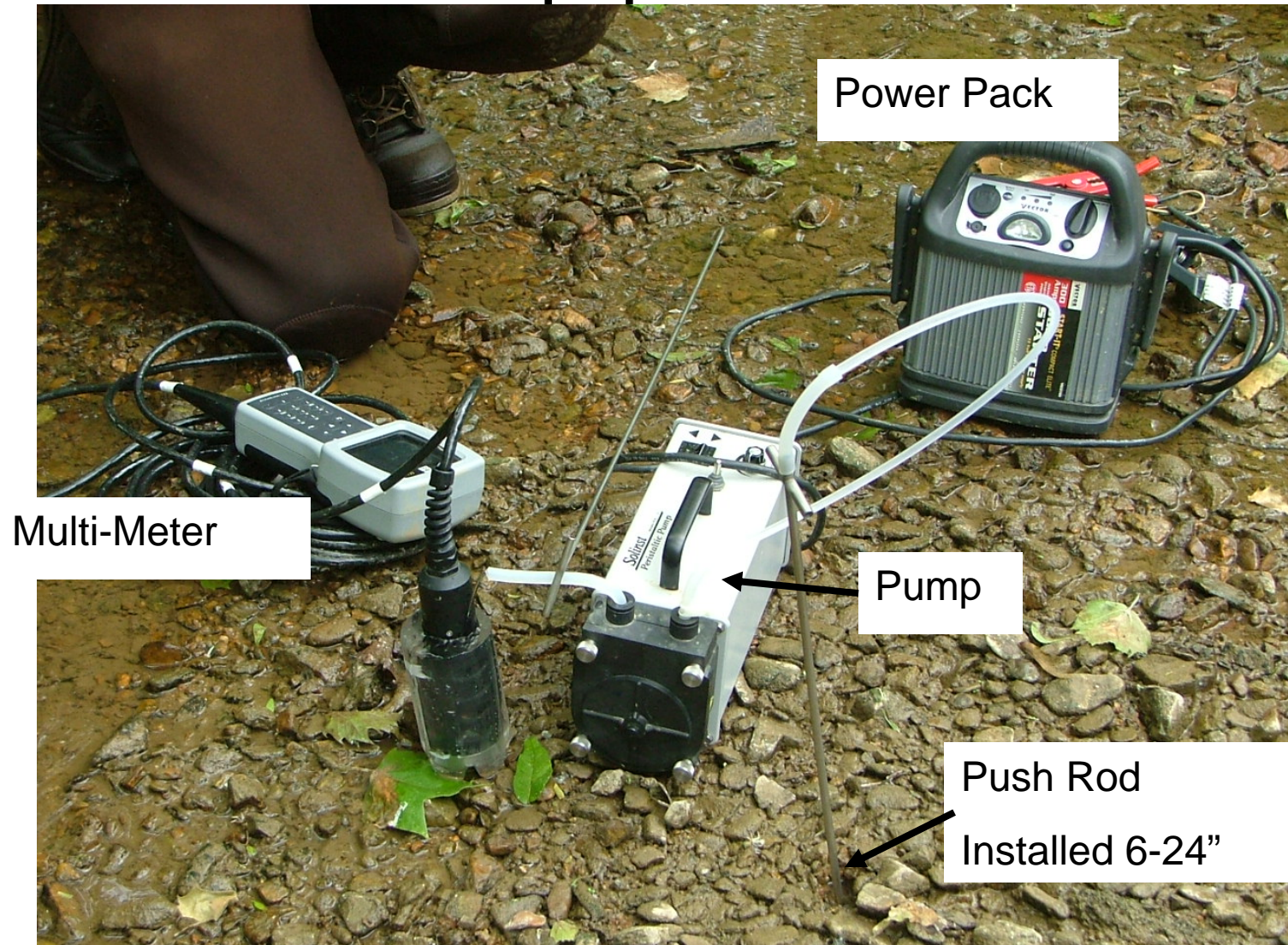
Bank Soil Characterization, R4

vertical channels between flags in ppm



Enough Hg in bank soil to account for Hg in sediment beneath bank

Pore/Hyporheic Water Sampling Equipment



Also measure water level in manometer relative to river water level

Porewater (GW) Pattern at Bank Dissolved (ng/L); Spec Cond ($\mu\text{S}/\text{cm}$) May 2006



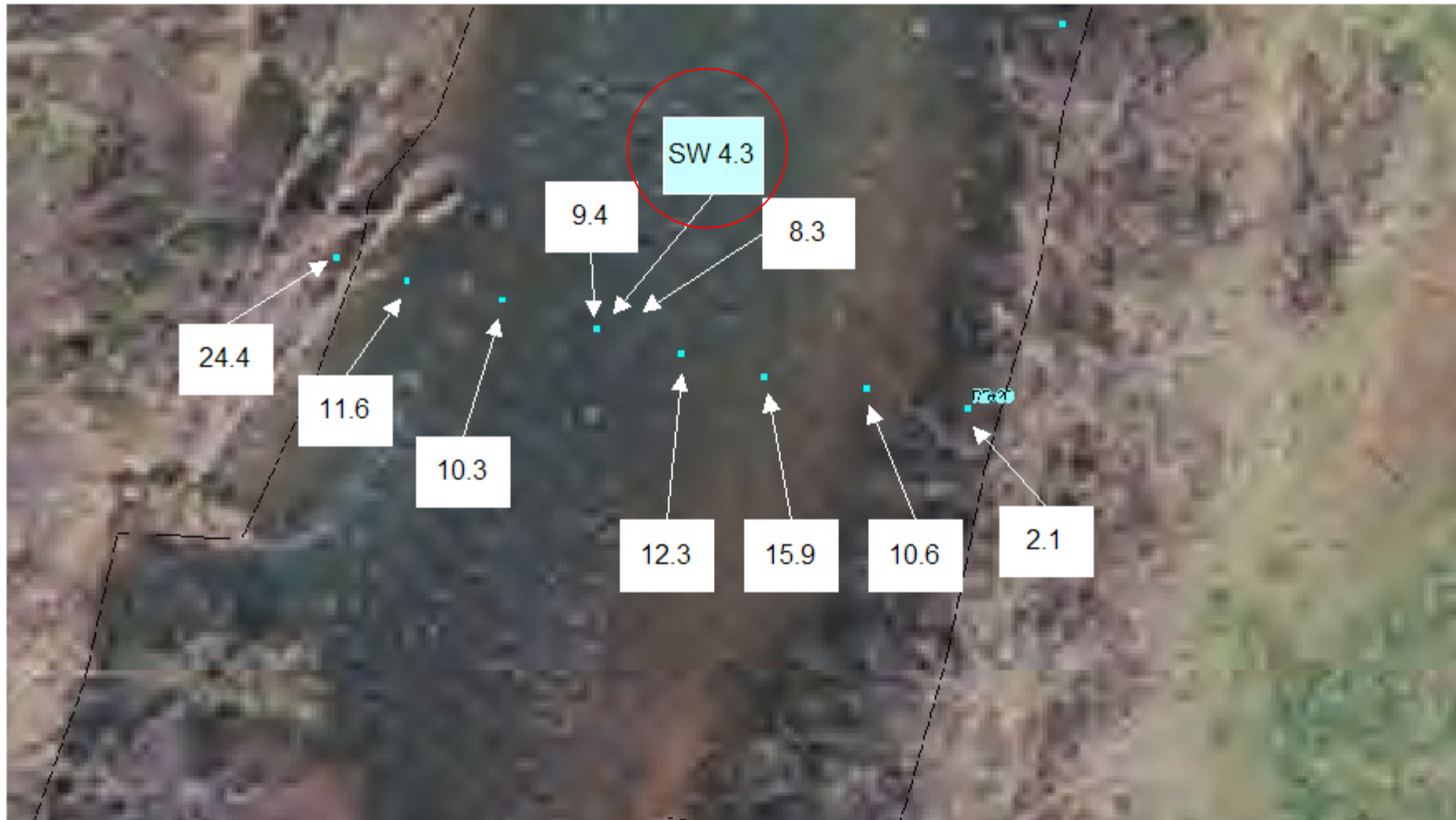
Hyporheic Water Pattern

July 2006, Including one SW



Hyporheic Water Transect

Dissolved THg (ng/L)-July 2006



[Hg] in streambed hyporheic zone generally 2x to 3x surface water (SW) value

Diffusion Buckets



Intended as a device to isolate a section of near-bank sediment from continuous “flushing” by upstream surface water, i.e., a simplified benthic flux chamber



Diffusion Buckets - July 06

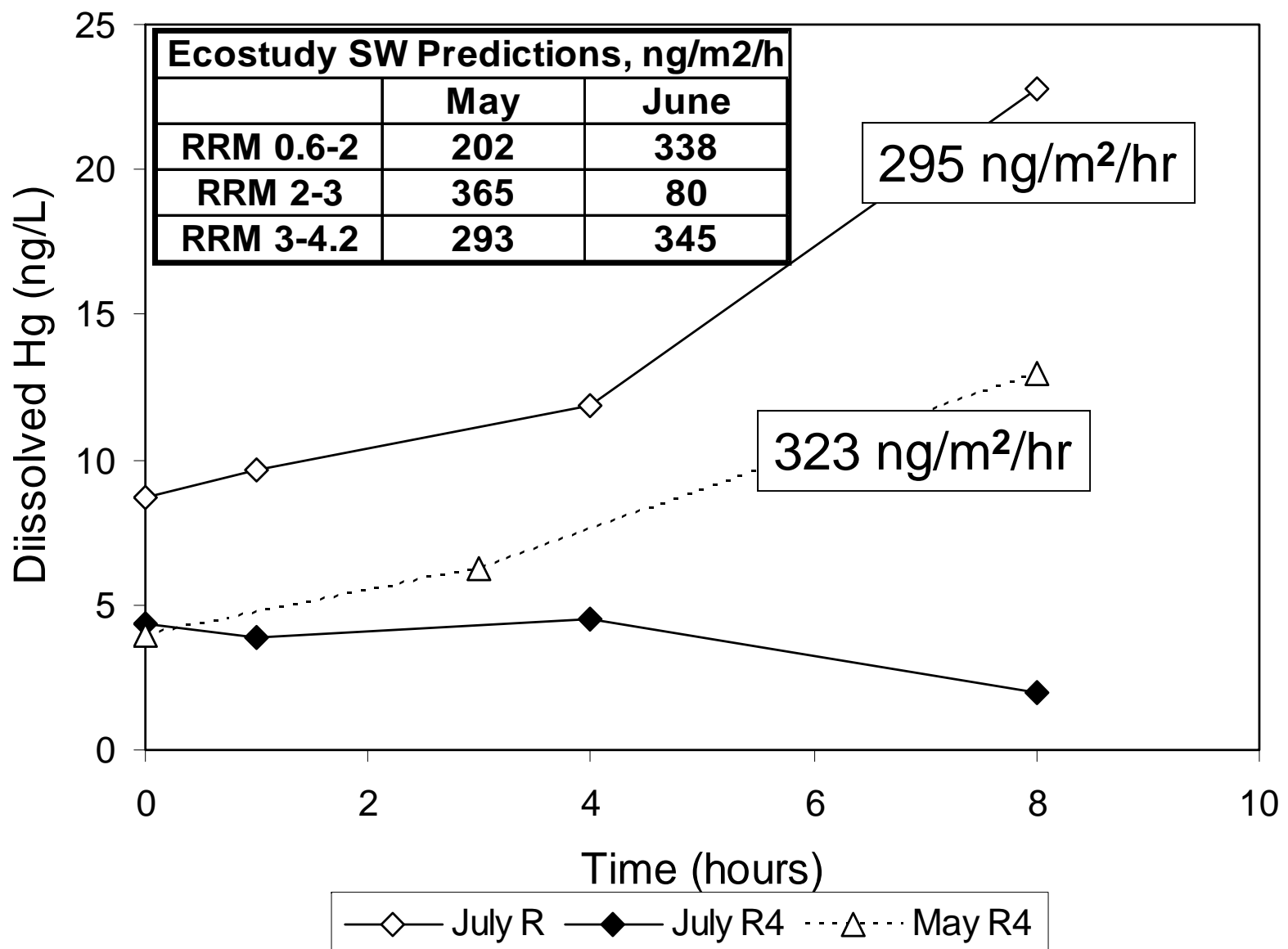
R-4



R



Diffusion Bucket Results



- Additional diffusion bucket data pending from 5 locations monitored for up to 23 hours in Sept 2006

Near Bank Results - May/July 06

- Strong SW “signal” at near-bank Basic Park location
- SW results seem to “follow” sediments
- Sediments perhaps more “localized” than previously expected?
- Sufficient bank soil Hg to account for sediments
- Shallow groundwater at base of bank relatively clean in most locations (i.e., lower than SW)
 - Pore water elevations do not “follow” SW or seds
- Near-bank sediments appear to release Hg at rates comparable to apparent “whole” river releases

Soil/Sediment Leaching Study

Objective: Determine whether Hg release from bank soils and near-bank sediments follows a “simple” desorption equilibrium.

Experimental Approach

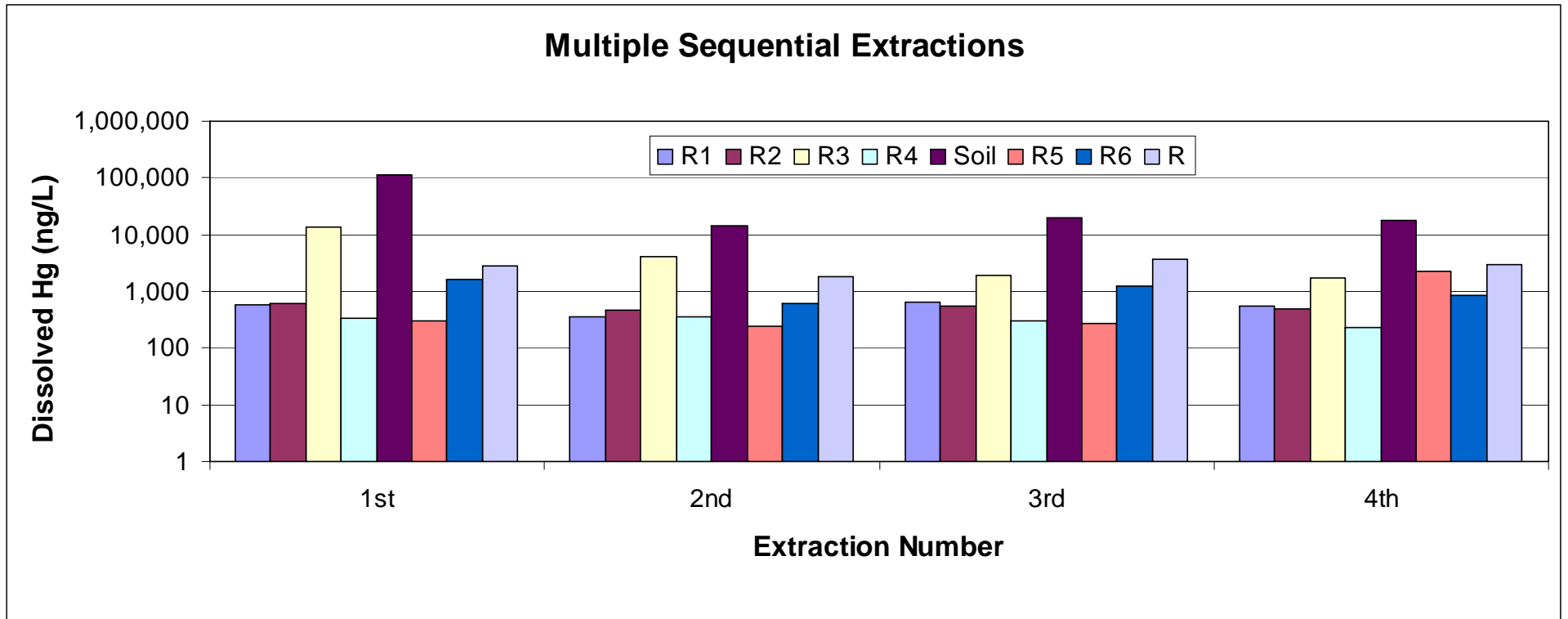
- Collect representative soil and sediments from study area at Basic Park.
- Perform four (4) successive extractions of each sample with DI water at solution/solid=10 (40 mL/4g)
- Analyze extracts for filtered (0.4 micron) mercury.
- Compare leaching patterns.

Possible Leaching Patterns

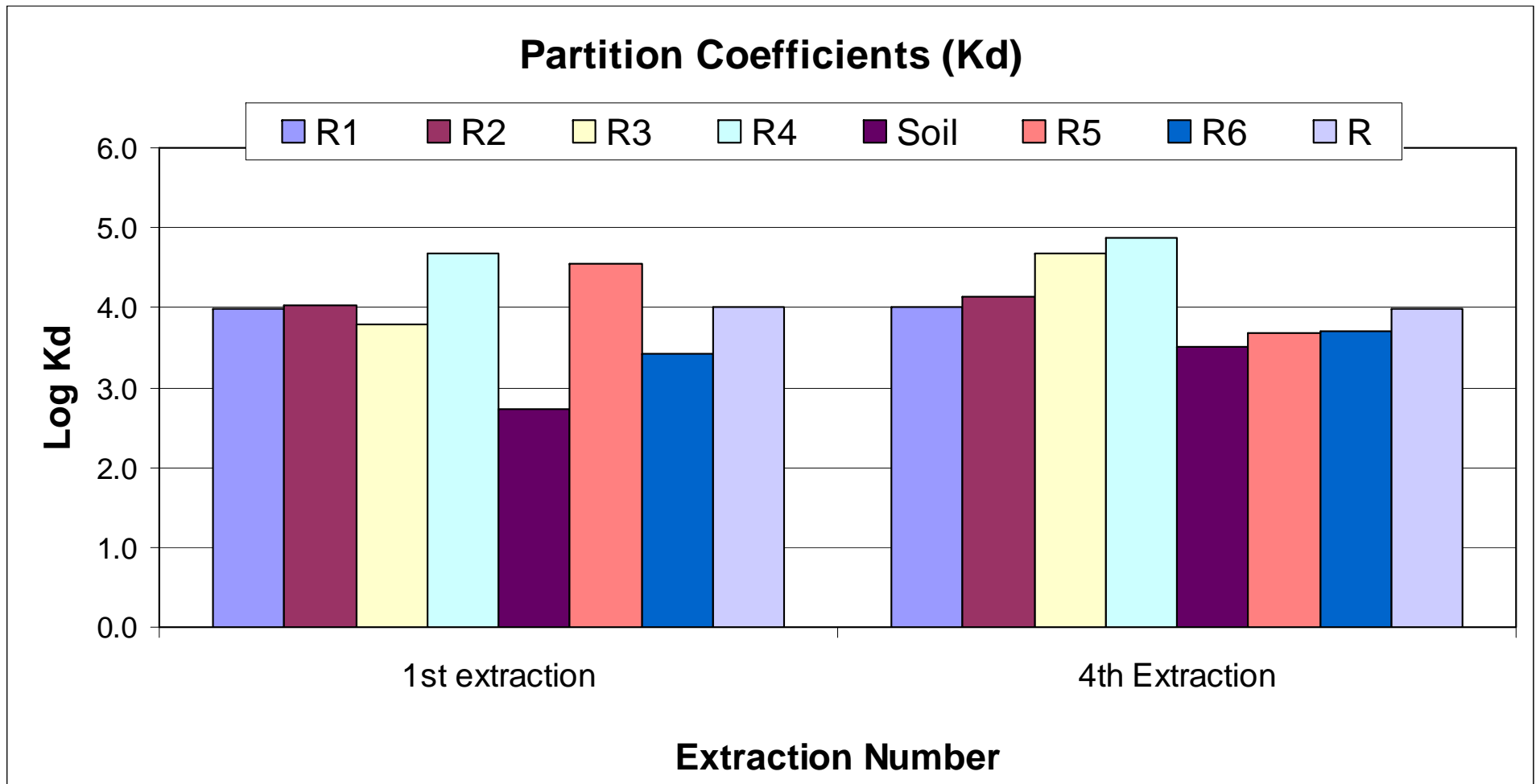
- Constant aqueous [Hg]
 - Suggests sorption/solubility “equilibrium” (infinite source)
- Decreasing aqueous [Hg]
 - Suggests “washout” of a highly soluble compound (e.g., HgO, HgCl₂)
- Increasing aqueous [Hg]
 - Suggests dissolution of an “occluding” compound or presence of a sparingly soluble compound (e.g., Hg₂O₂, Hg₂Cl₂, HgS) affected by redox reaction(s)

Soil-Sediment Samples

Sample ID	Total Hg (mg/kg dw)
R1	5.64
R2	6.57
R3	81.6
R	28.4
R4 - Soil	59.6
R4	16.6
R5	10.5
R6	4.36



Similar aqueous [Hg] across all four extractions.
 Bank soil produced highest aqueous [Hg]



Log Kds much lower than observed for extractions run at higher solution/solid ratios.

Is this simply a “particle effect” due to overestimation of true aqueous [Hg] due to presence of colloidal Hg?

Conclusions-Leaching Study

- Leaching patterns (constant [Hg]) suggest all samples have Hg in “exchange” positions.
- Bank soil produced the lowest partition coefficients (K_d s) especially at the initial extraction.
- Release and diffusion of Hg from these materials could possibly account for longitudinal and transverse patterns in South River surface water Hg.
- Results from Rob Mason (shake & bake), and an earlier simulated TSS leaching, showed much lower releases but were conducted at much higher solution/solid ratios.

Path Forward-Leaching Study

- Verify high aqueous [Hg] associated with the sediment samples by collecting “shallow” porewater samples from each location.
- Repeat extraction of one sample from set with filtered river water from SR-01 (Lyndhurst).
- Characterize the physical/chemical nature of Hg in these kinds of leachates (e.g., volatility, molecular weight, reactivity)