

# Mechanistic Studies and Tool Development

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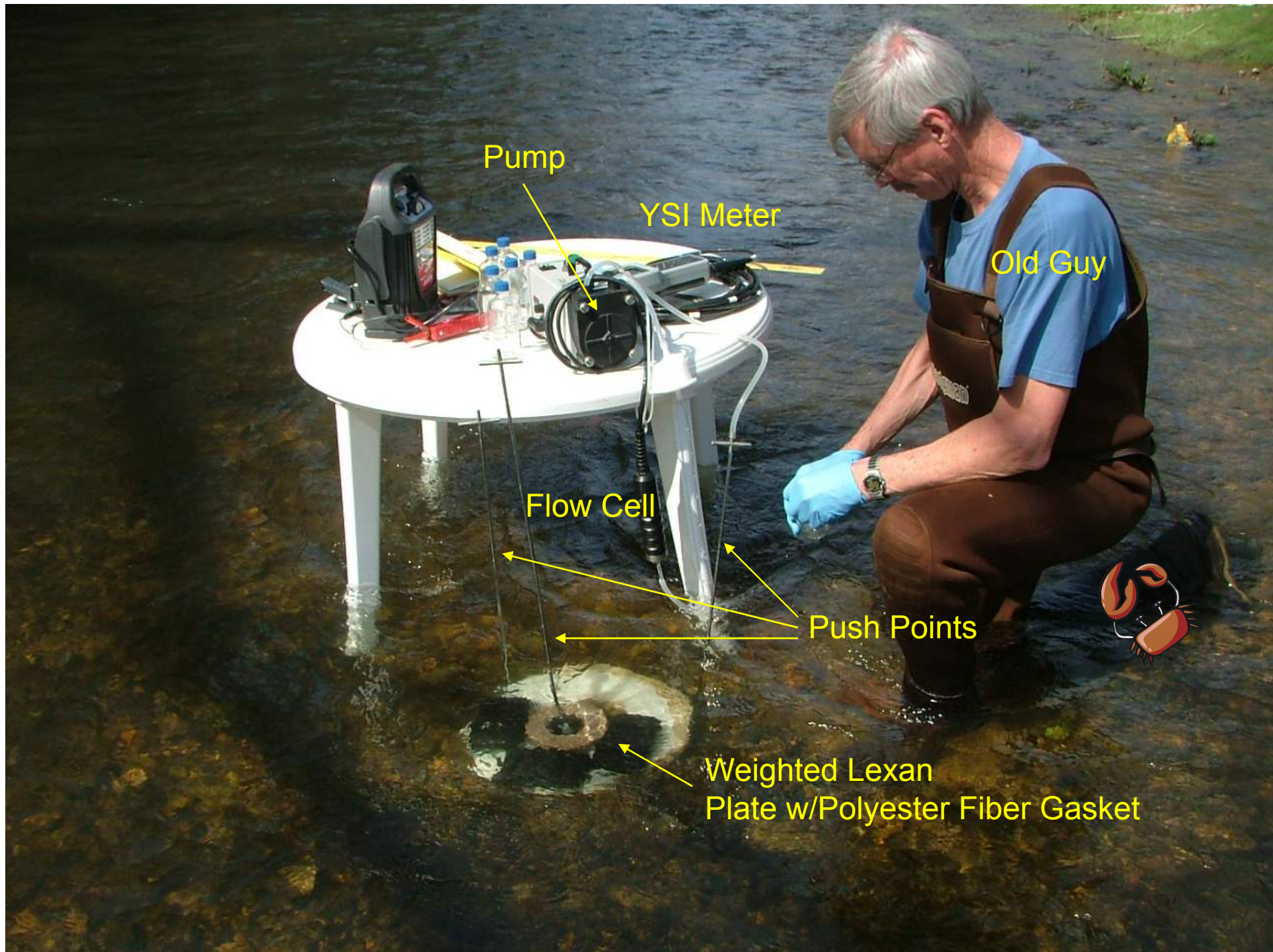
October 2008

# 2008 Focus Areas

- Gravel Bar Hyporheic Zone Characterization
- Characterization of “reactive Hg” [Hg(II)] in surface water and outfalls.
- Methyl Hg Desorption Kinetics

# Hyporheic Sampling at Basic Park Gravel Bar

- Objective was to retrieve water and sediment samples at increasing depths in the bed
  - Reduce/eliminate influence of river flow on near-interface water samples using a “gasket”
  - Sediment collected by “sequential excavation” and “micro-guzzling” (April and June only)
  - Water collected April, June and September.
  - Measured hydraulic gradients manometrically
- Persistent Question Addressed: Are Embedded Gravels Zones of MeHg Production/Release?



Pump

YSI Meter

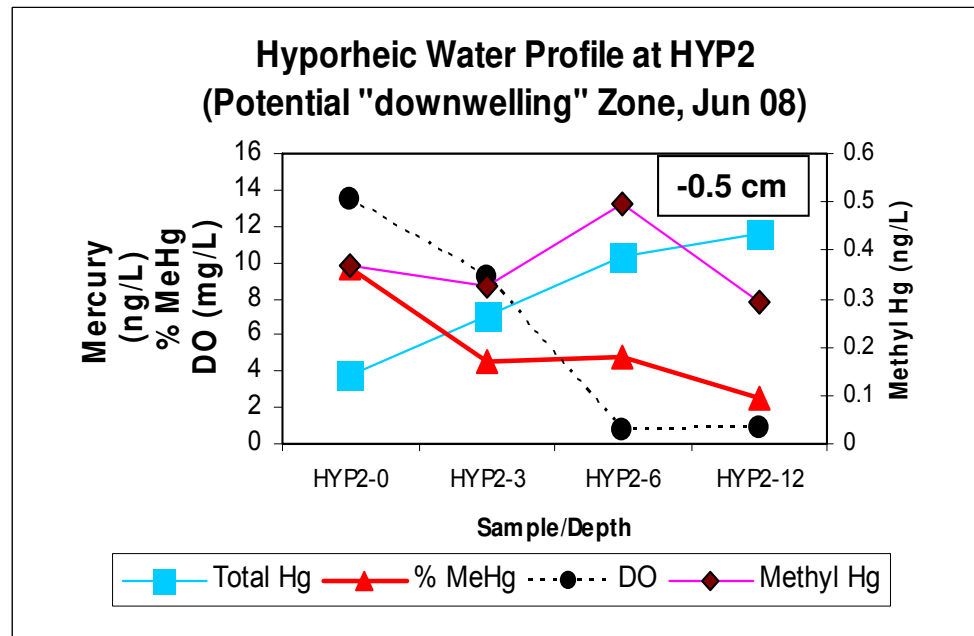
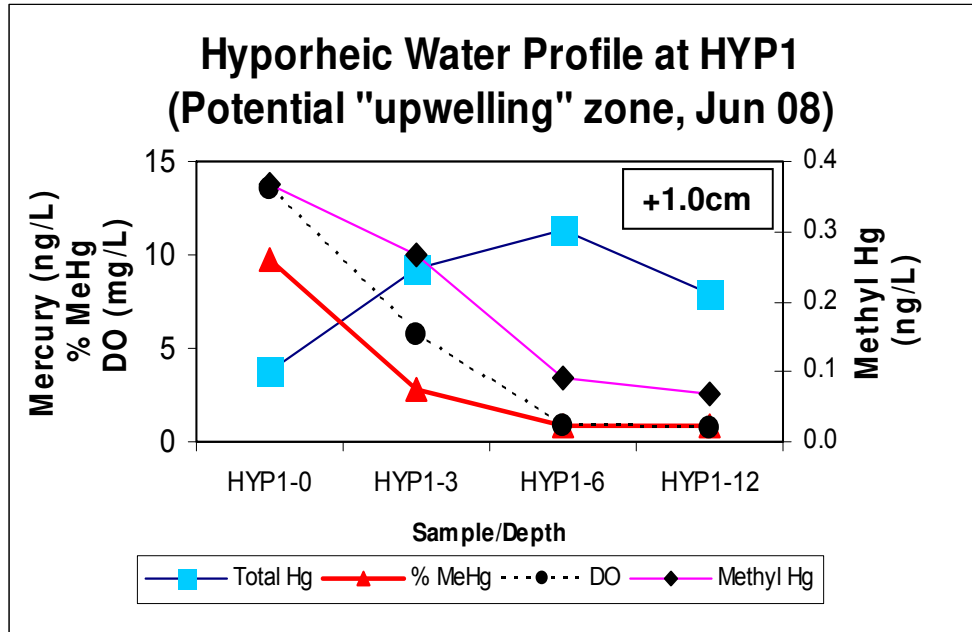
Old Guy

Flow Cell

Push Points

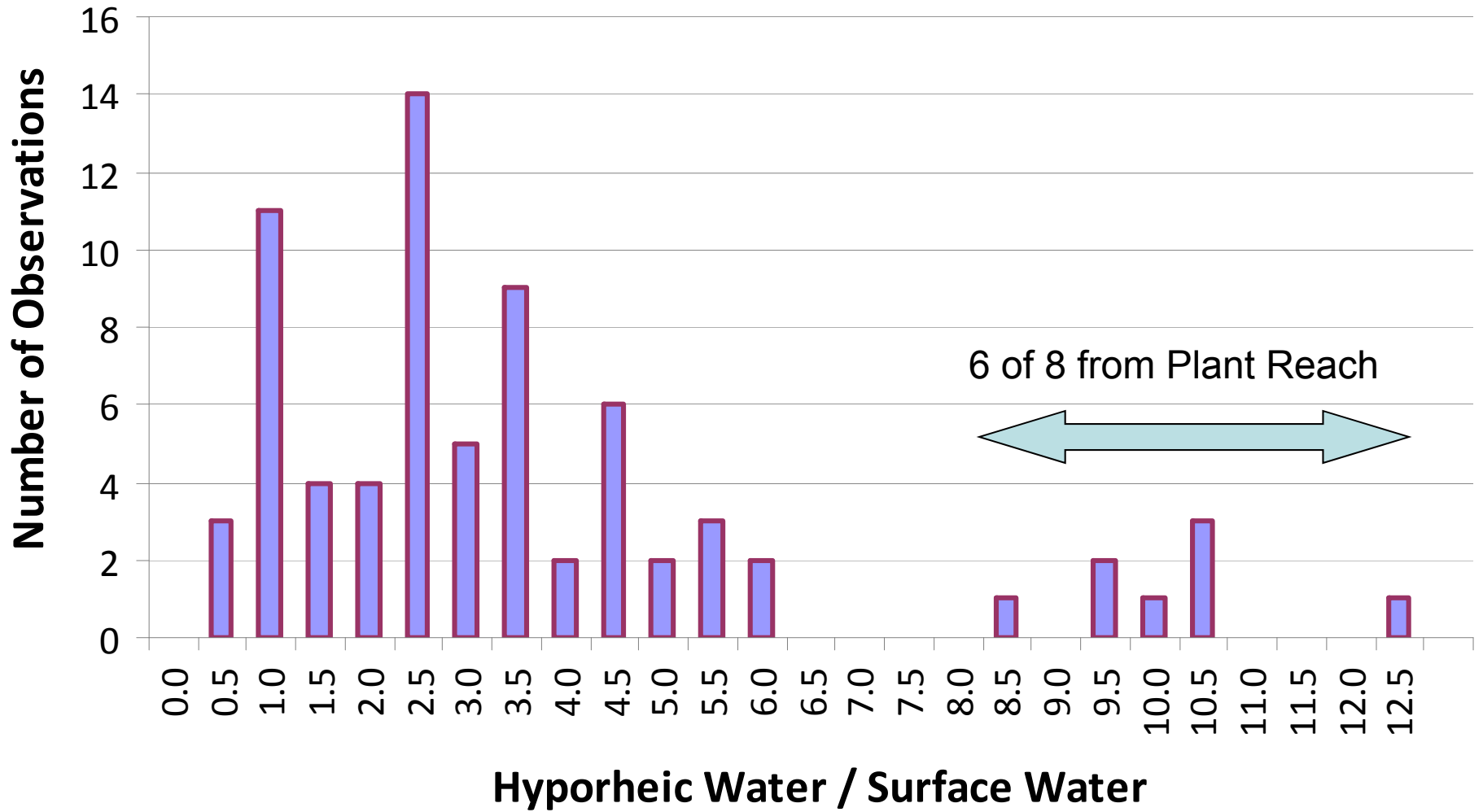
Weighted Lexan  
Plate w/Polyester Fiber Gasket



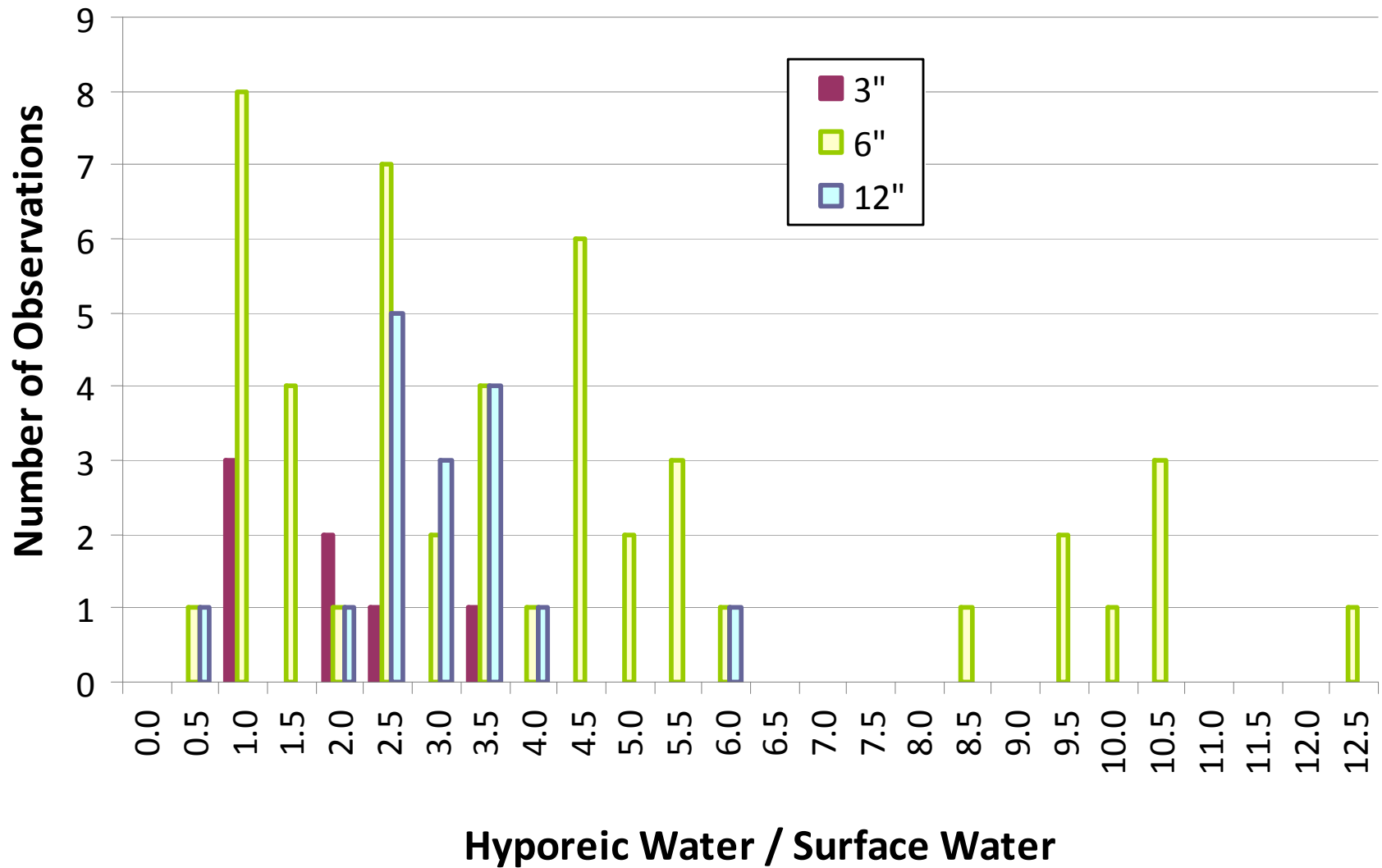


# Hyporheic/Surface Water Concentration Ratio Total Hg - All Depths/Locations

Increasing Diffusive Driving Force



# Hyporheic/Surface Water Concentration Ratio Total Hg - By Depth







# Stratified Guzzling Layer Prep



## Sediment Profiles at Hyporheic Stations

Station	Depth (Inches)	Total Hg (ng/g)	Methyl Hg (ng/g)	%Methyl Hg	%LOI	MeHg-LOI (ng/g)
HYP2	1 to 3	9764	13.86	0.14	19.0	73.0
	3 to 6	14381	5.60	0.04	13.4	41.8
	6 to 9	13119	2.68	0.02	7.87	34.1
HYP3	1 to 3	8432	6.90	0.08	8.40	82.1
	3 to 6	7248	5.83	0.08	10.0	58.4
	6 to 9	12104	4.24	0.04	11.6	36.6

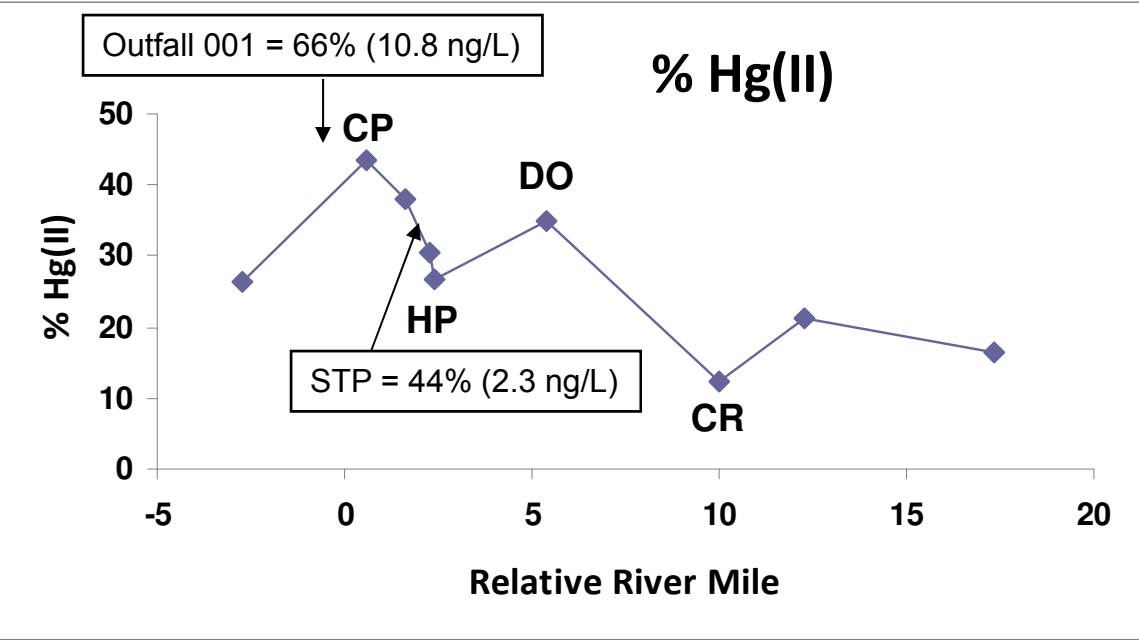
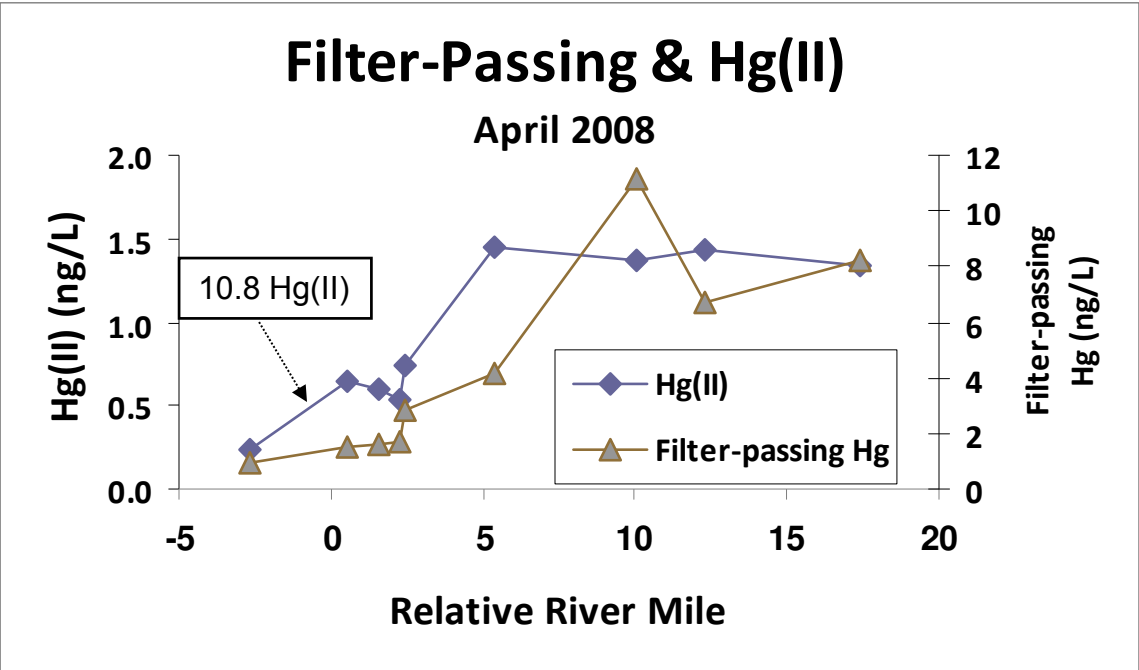
- THg increases (slightly) with depth
- MeHg, MeHg-LOI and %MeHg decrease with depth
- MeHg atypically low (too early in spring??)

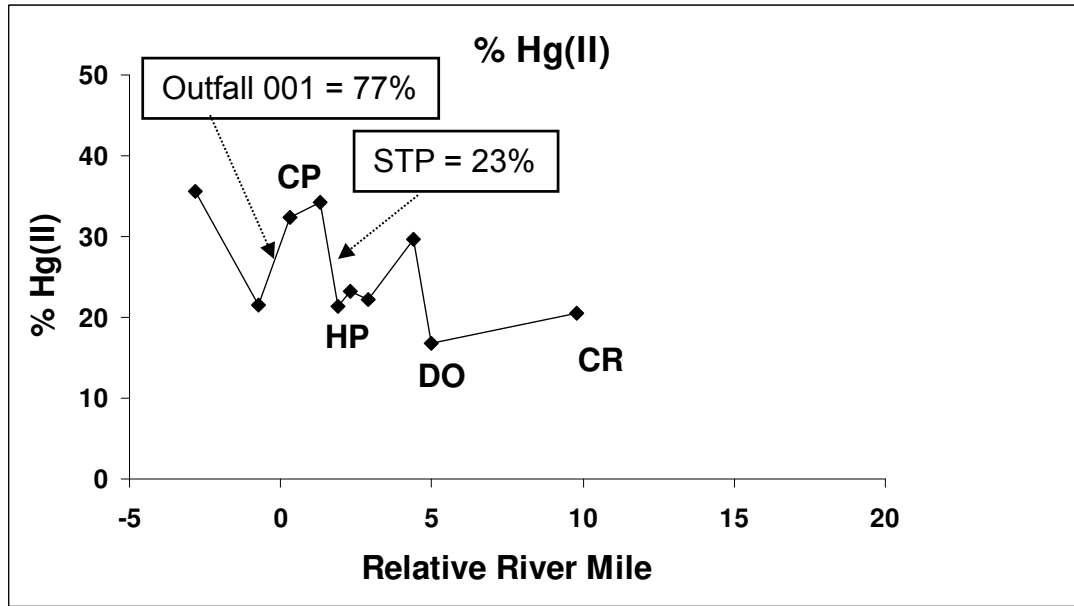
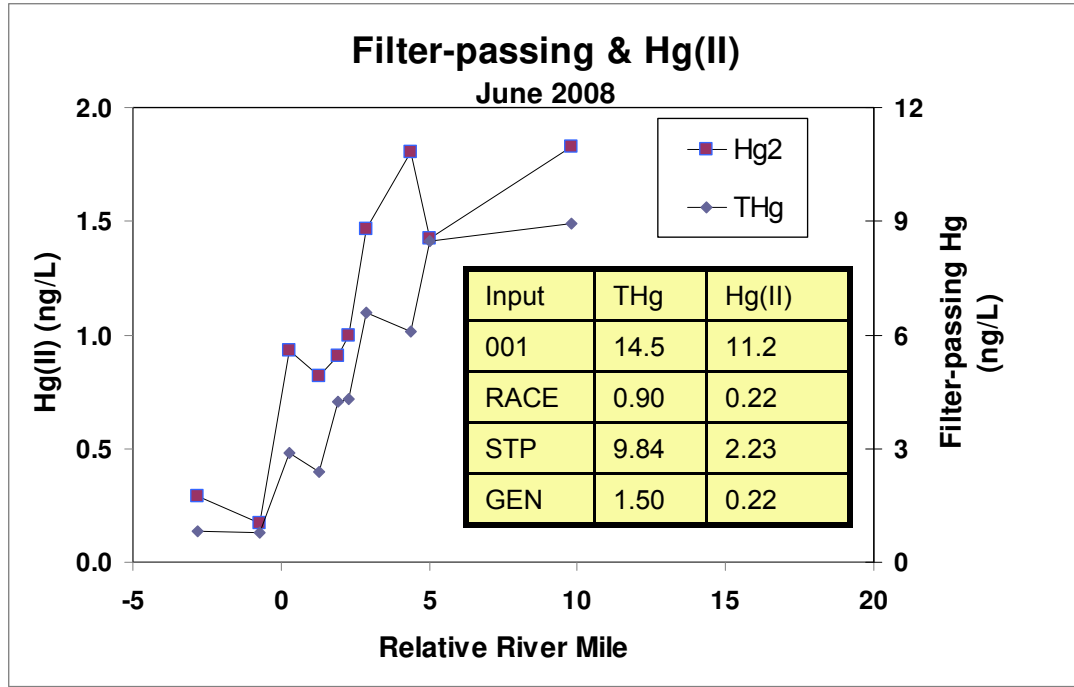
# Stratified Hyporheic Indications

- THg & MHg behaviors opposite
  - THg peak deeper
  - MHg peak shallower
  - Both for sediment and pore water
- Pattern suggests local MHg production near interface, vs arrival sorbed to same solids, but....*higher [MHg] is not where dissolved oxygen lowest*

# Reactive Mercury [Hg(II)]

- Potential as surrogate for “bioavailable”
  - May be “pathfinder” for inputs of “new” Hg
- Reactive=Easily reducible to Hg(0)
  - Likely inorganically complexed Hg(II), incl Hg<sup>2+</sup>
  - Highly bioavailable (but some debate)
- Can we use the Lumex to measure in “field”, i.e., “close interval sampling”?
  - Trial setup in Waynesboro office on April 5
- Trial surveys from SR01 to Harriston on April 8, and SR01 to Crimora on June 24
- Hg(II) samples sent to Studio Geochimica.





# Hg(II) Indications

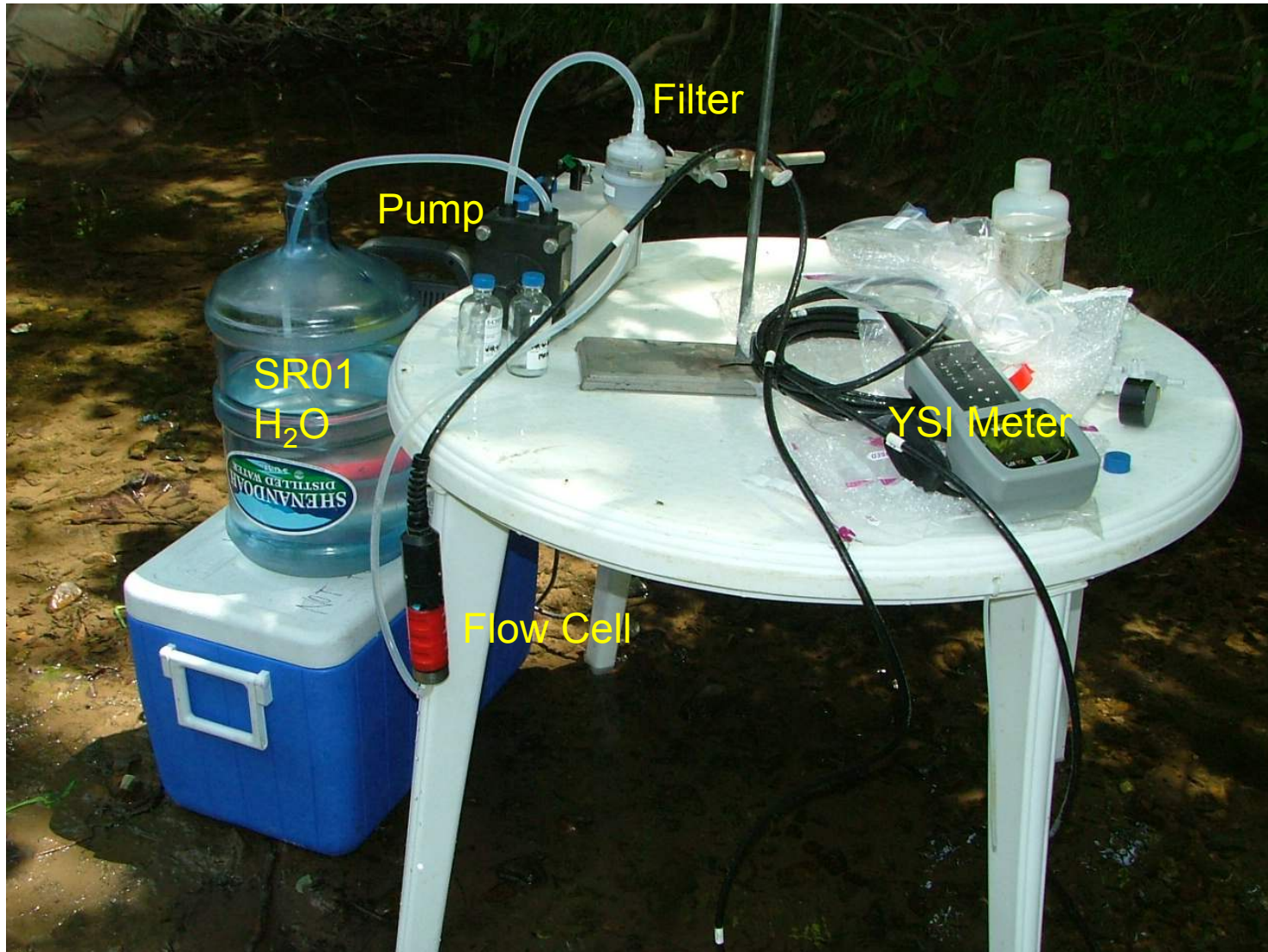
- [Hg(II)] generally tracks filter-passing Hg and increases downstream.
- % Hg(II) generally decreases downstream.
- Unlikely to be very useful as an input pathfinder: *Inputs not significantly “enriched” in Hg(II) compared with river surface water.*

# Methyl Hg Desorption

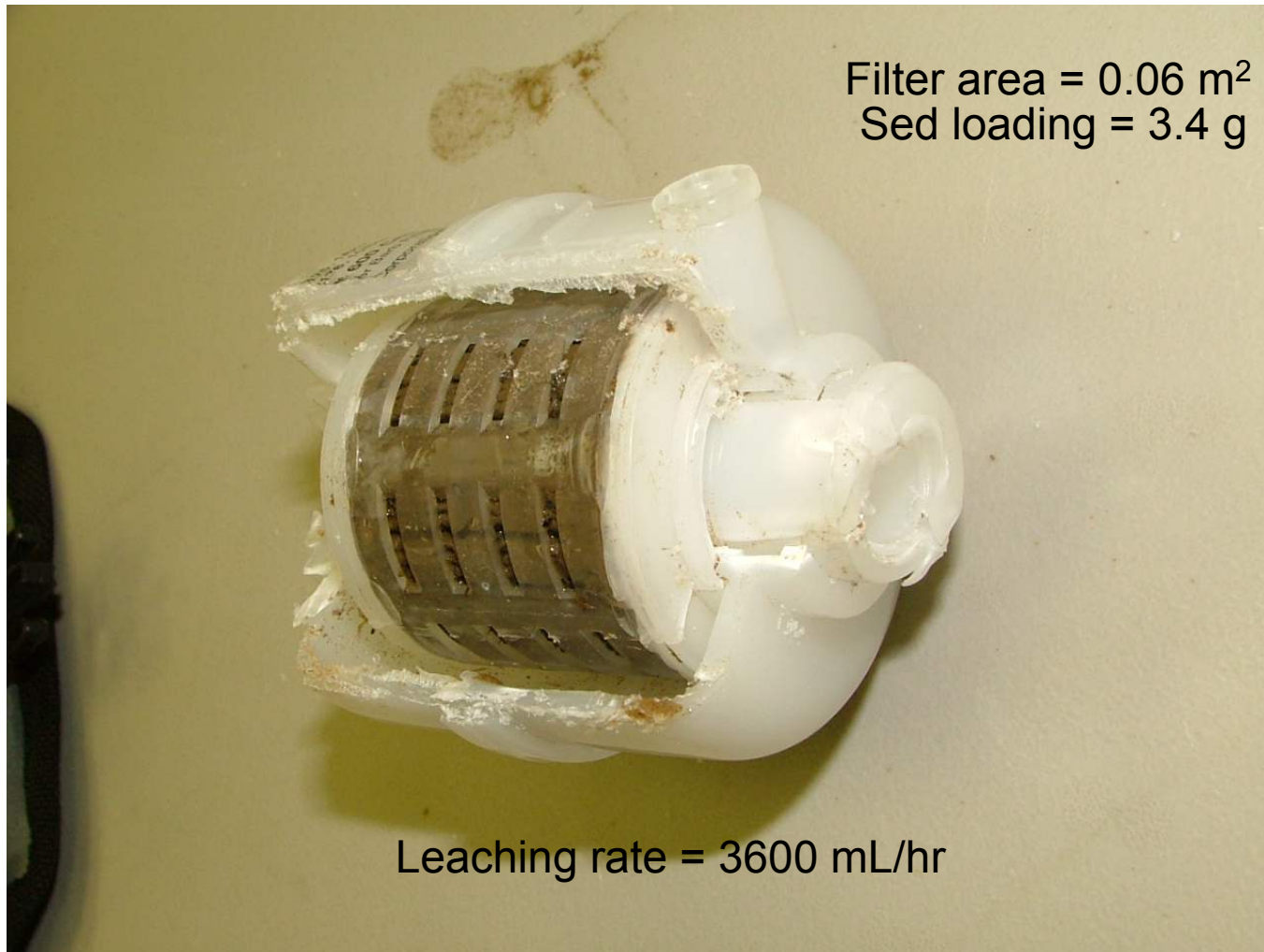
- What role does desorption of “pre-formed” MHg from sediments play in observed downstream increase in [MHg] in sfc water and rise in [MHg] in BFCs?
  - Partition coefficients for MHg lower than than for inorganic Hg but MHg still predominantly particle-associated.
  - Once formed could supply sfc water with filter-passing MHg even when favorable methylation conditions are absent.
  - Need to characterize kinetics of desorption vs methylation



# MHg Filter Loading Extractions

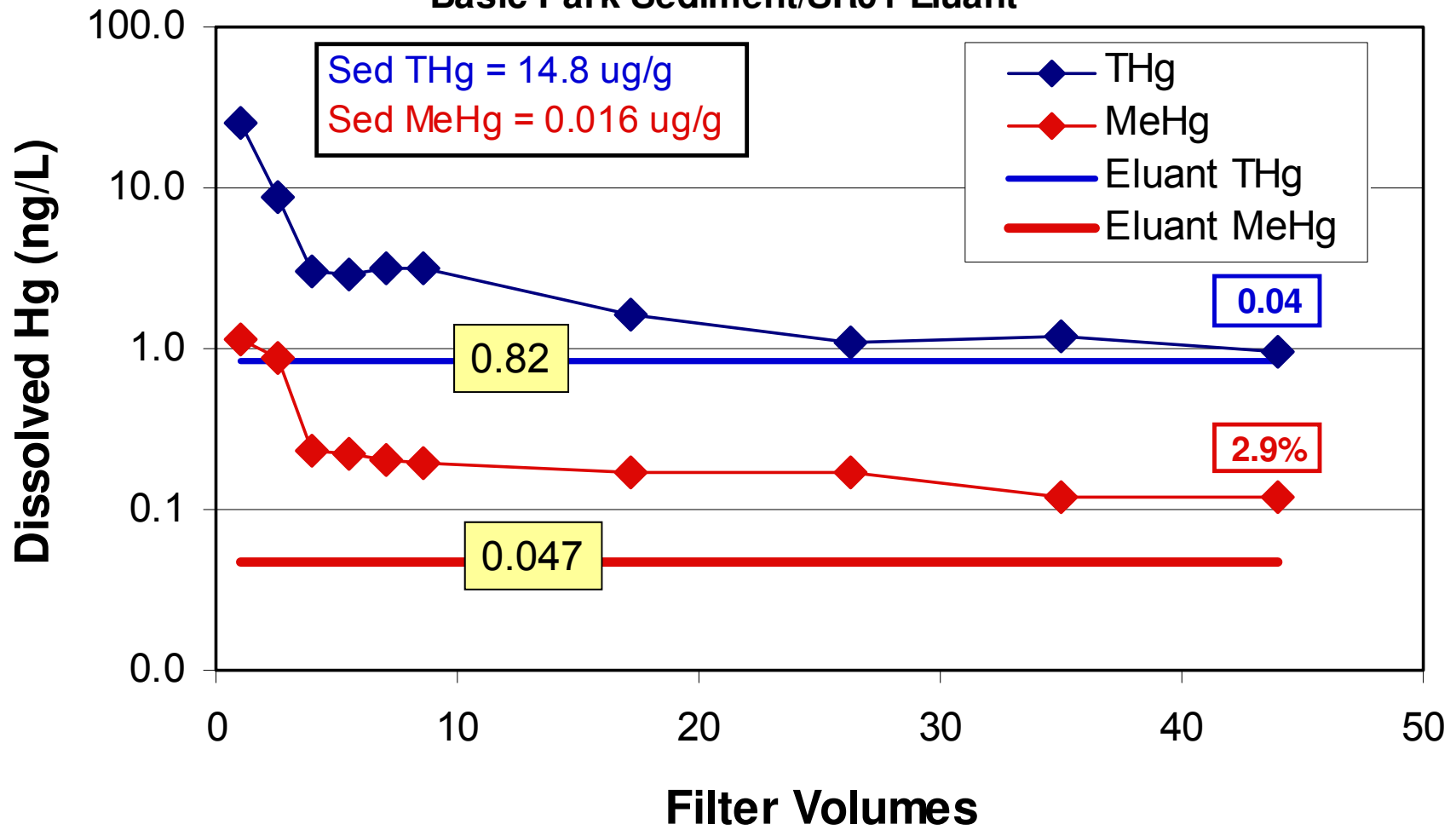


# Filter Cut-away Showing Sediment Layer



# Sediment Desorption

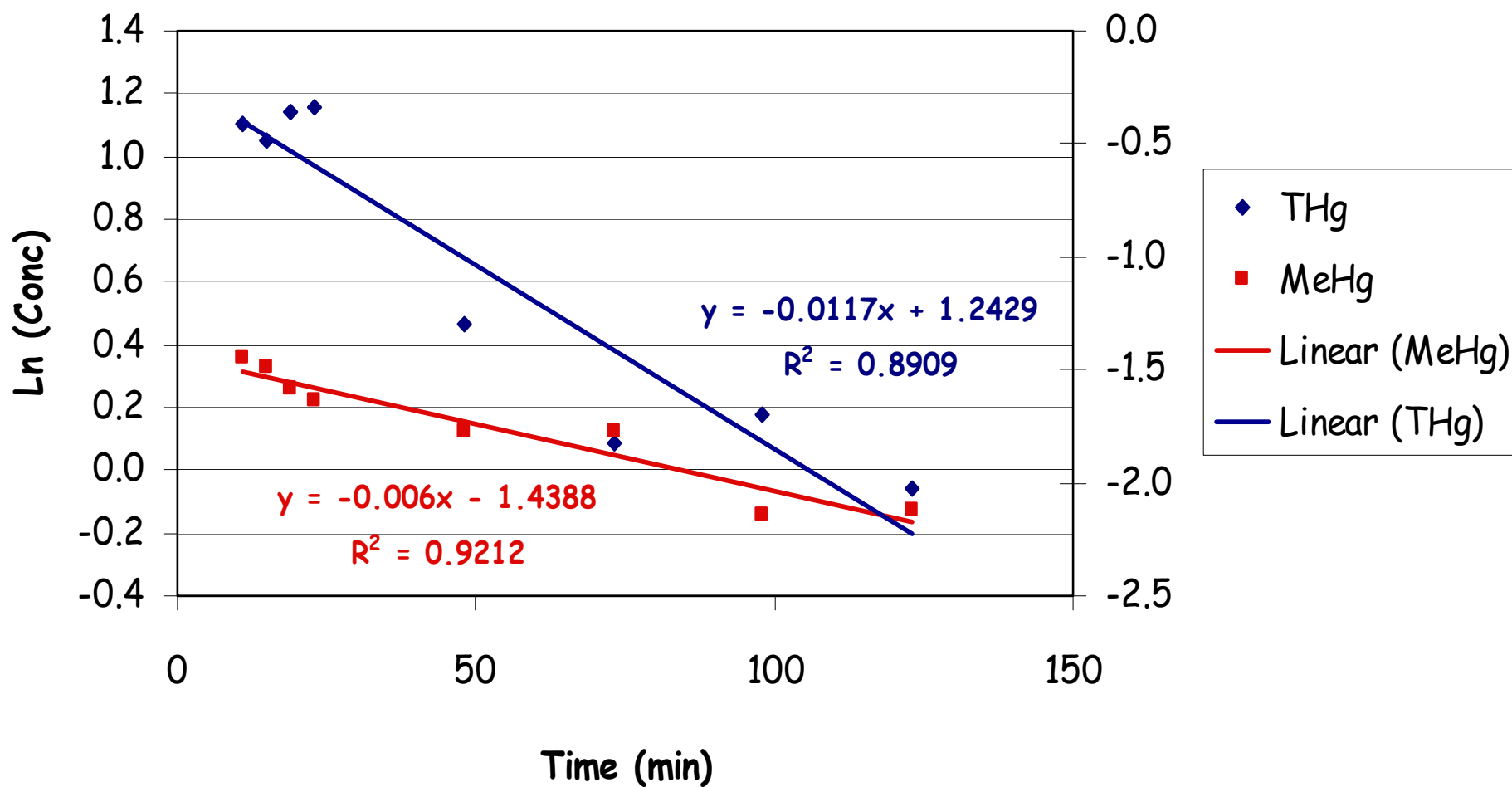
Basic Park Sediment/SR01 Eluant



# Methodology-Kinetic Analysis

- Consider both mechanistic and empirical kinetic models
  - mechanistic rate laws
    - zero-, first-, and second-order chemical kinetics
  - Parabolic diffusion equation
    - diffusion-controlled phenomena are rate limiting
  - Empirical equations
    - Elovich equation (heterogeneous chemisorption model)
    - Power function
- Only considered data for  $\geq 4$  filter volumes

# Check for First-Order Reaction

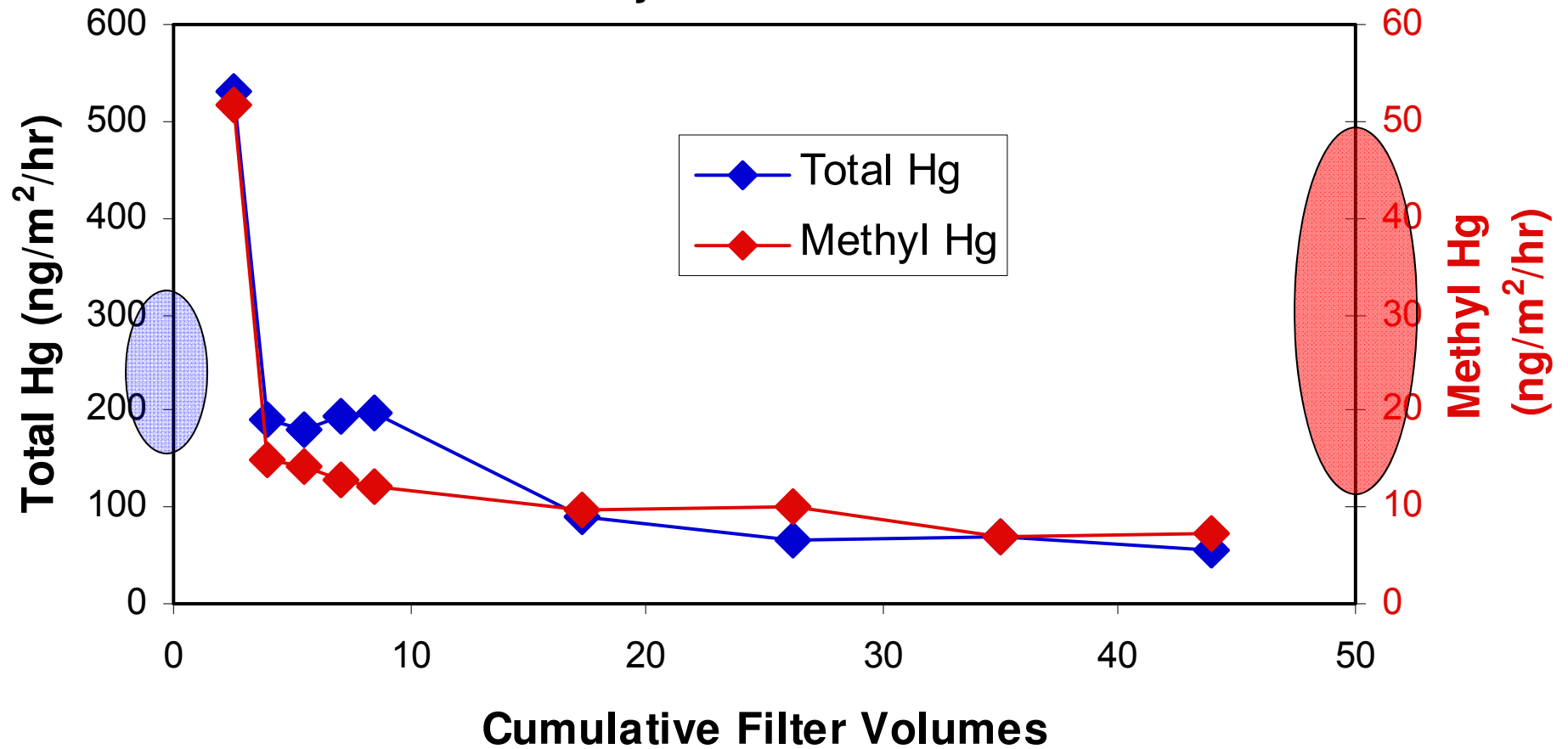


# Conclusions

- First Order rate Model
  - $k_1 = 0.0117 \text{ min}^{-1}$  ( $16.9 \text{ d}^{-1}$ ) for FIHg
  - $k_1 = 0.006 \text{ min}^{-1}$  ( $8.6 \text{ d}^{-1}$ ) for FMHg
  - **$t_{1/2} = 1 \text{ hr}$  for FIHg**
  - **$t_{1/2} = 2 \text{ hr}$  for FMeHG**
  - Compare to typical methylation and demethylation rates
    - $1\text{E-}05$  to  $5\text{E-}02 \text{ d}^{-1}$  in sediment and water column
  - These results suggest that desorption rates using background river water may be much faster than typical methylation and demethylation rates

# "Extractive" Hg Flux - BP Filter Sediment

Ovals = Jan-May 2005 Plant to Dooms Fluxes



# Results Pending

(September 2008 studies)

- Repeated elution w/BPK sediment:
  - Free water removed from filter at start
  - First elution with BPK water then elution with SR01
  - Stop/start flow with both eluants
- Adsorption experiment with clean sediment and contaminated river water



# Methyl Hg Desorption Indications (so far)

- Loaded filter – column leaching approach easy/fast to set up and operate
- Easy to impose “manipulations” like stopped flow, alternate eluants
- DO in freshly recovered slurries similar to in situ values (3-inch hyporheic)
- May be more a realistic “simulation” of river water flowing over sediment than at first glance
- Assumed to reflect predominantly desorption based on rates but this aspect needs further verification.