

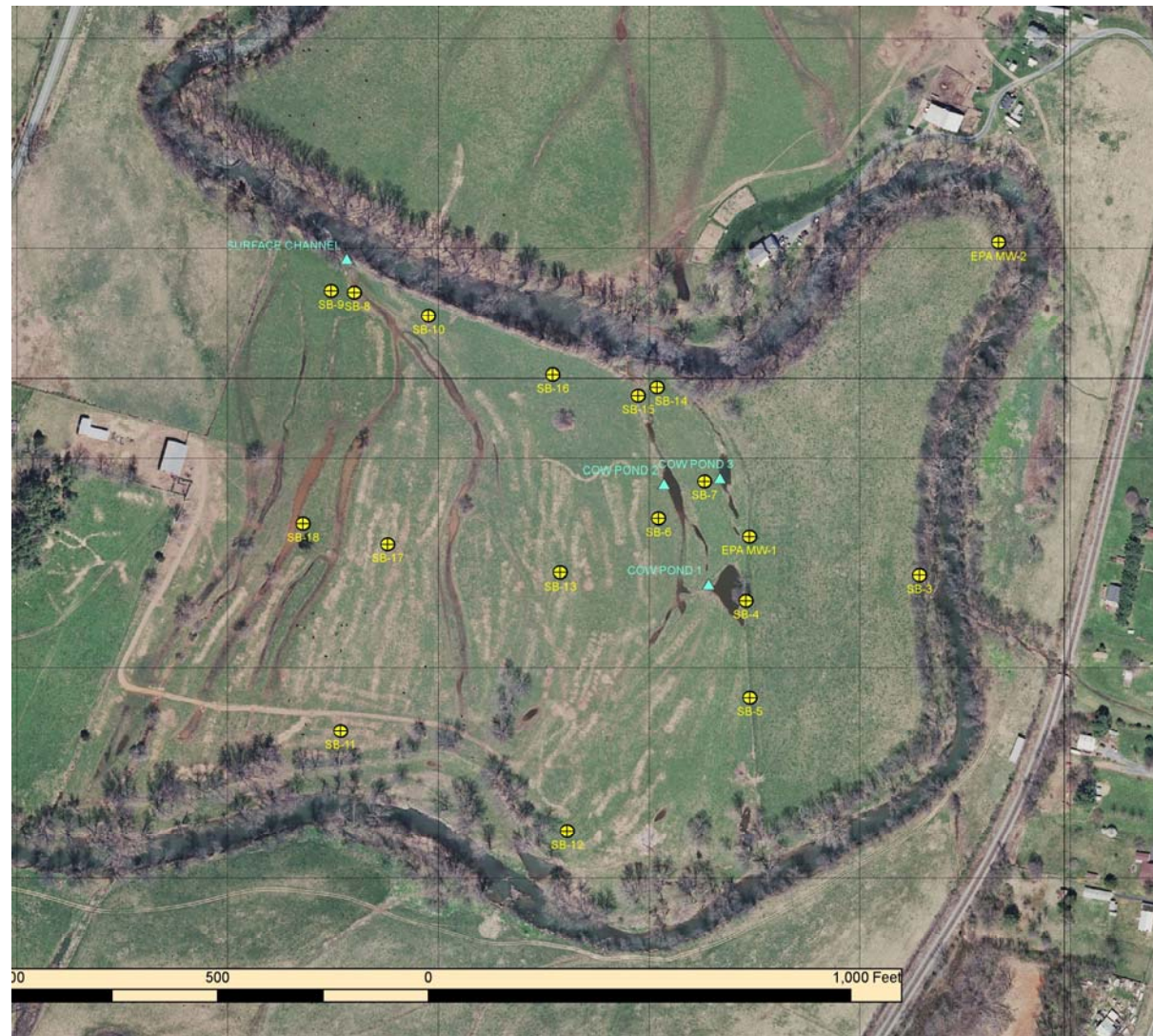
# Shifflet Farm – Soil & Groundwater Results, Part II

Mark Chappell, US Army Corps of Engineers, Vicksburg, MS

Joel Hennessy, US Environmental Protection Agency, Philadelphia, PA

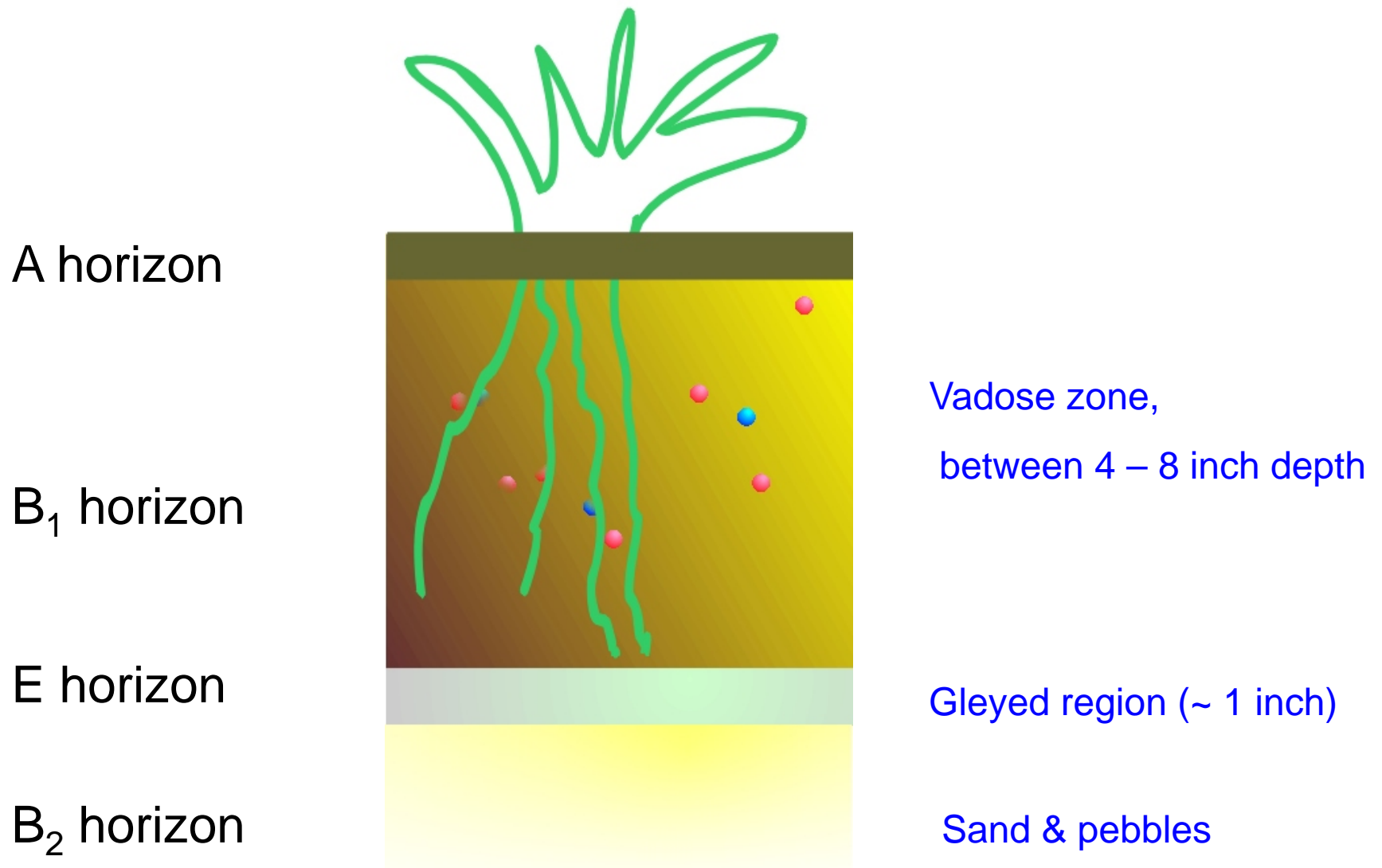


# Shifflet meander





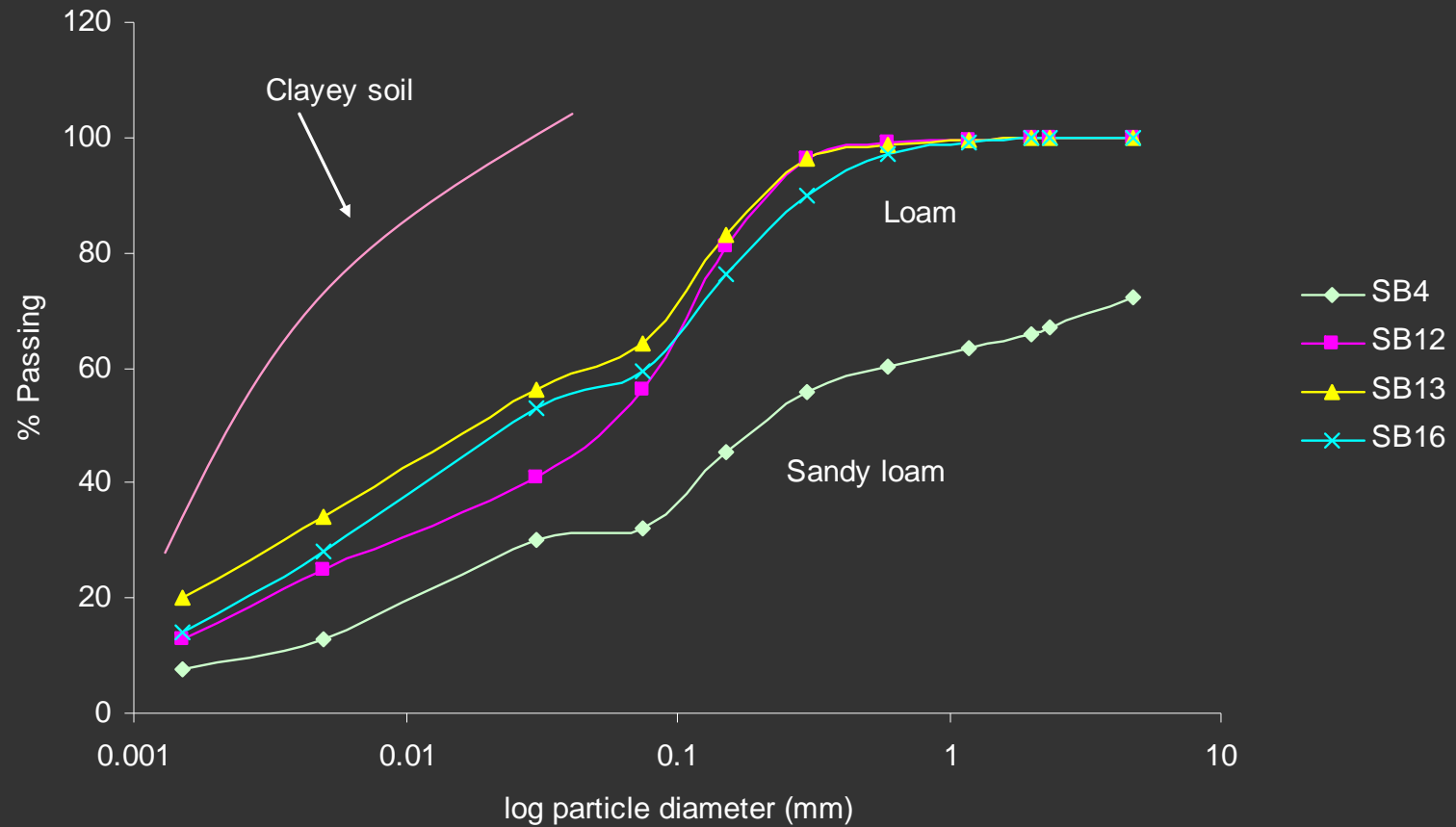
# Generalized soil profile



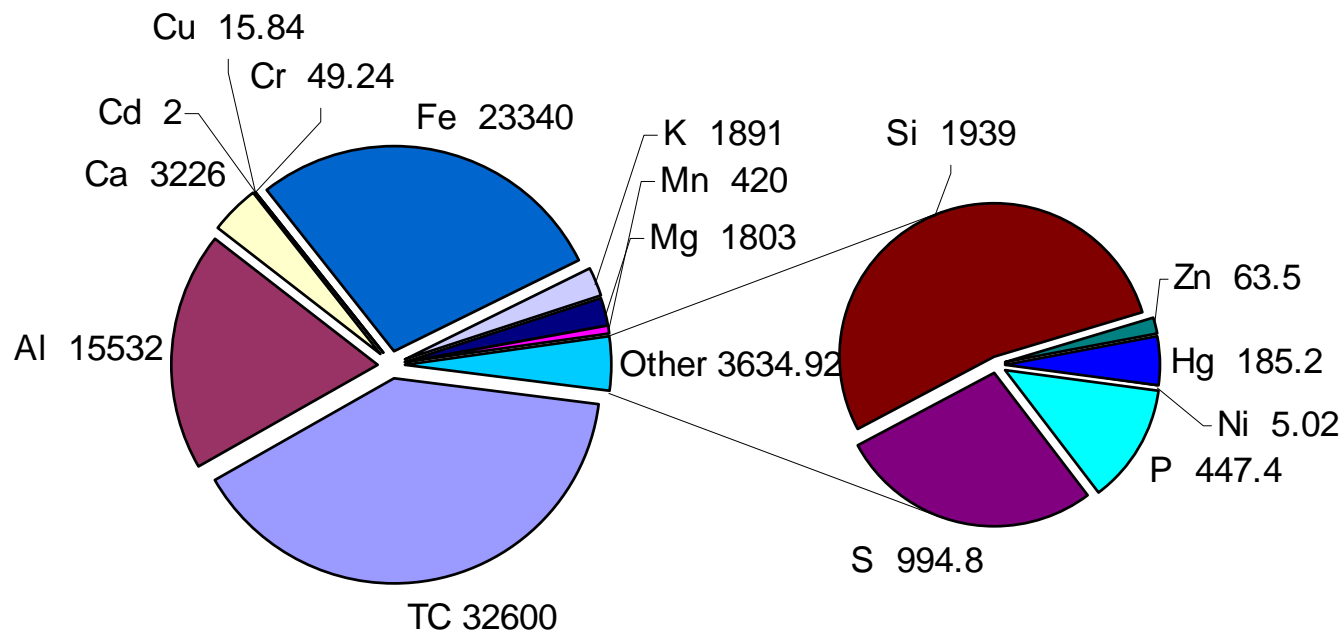
# Soils description



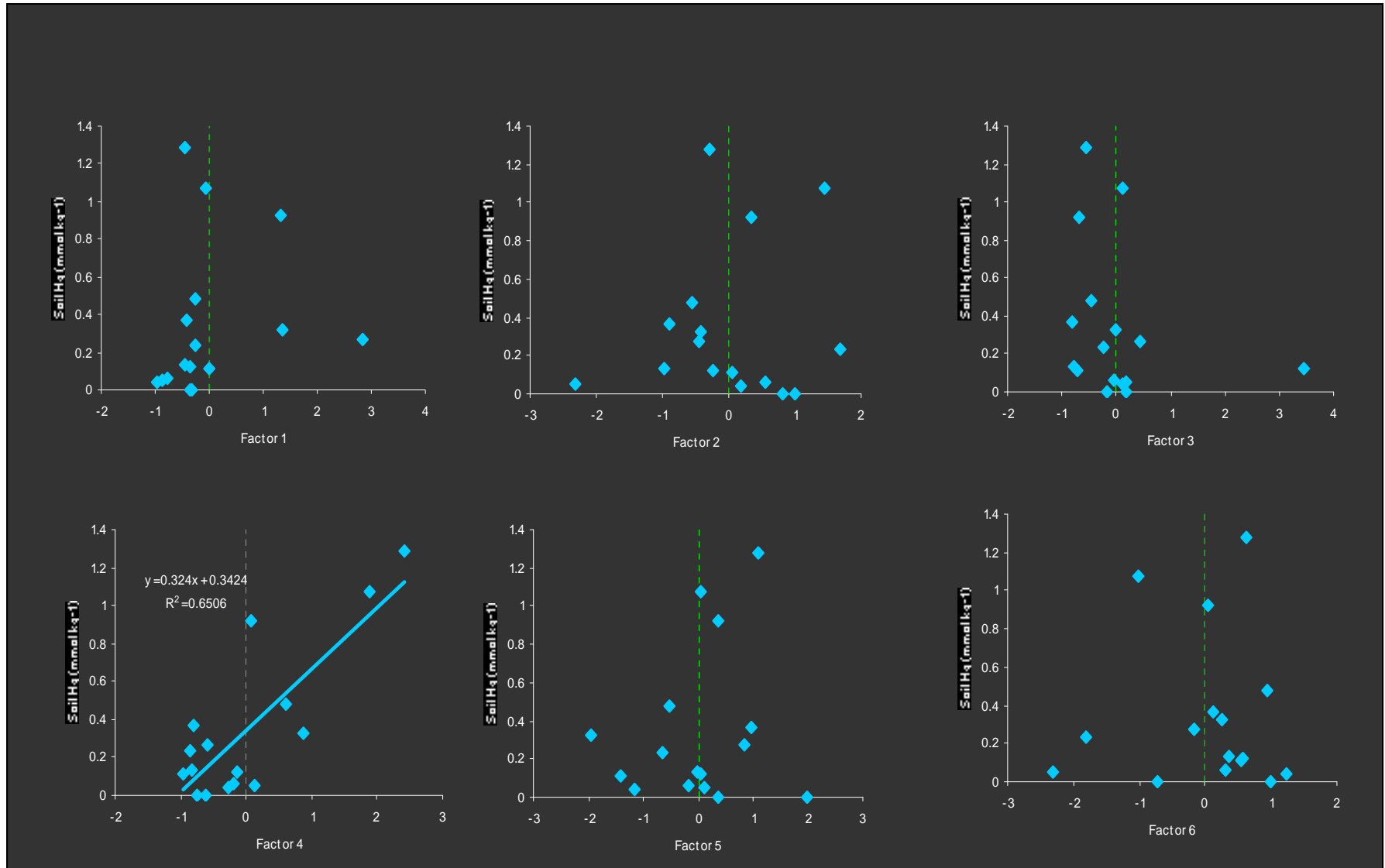
# Soil texture



# Chemical characterization



# Principal factor analysis

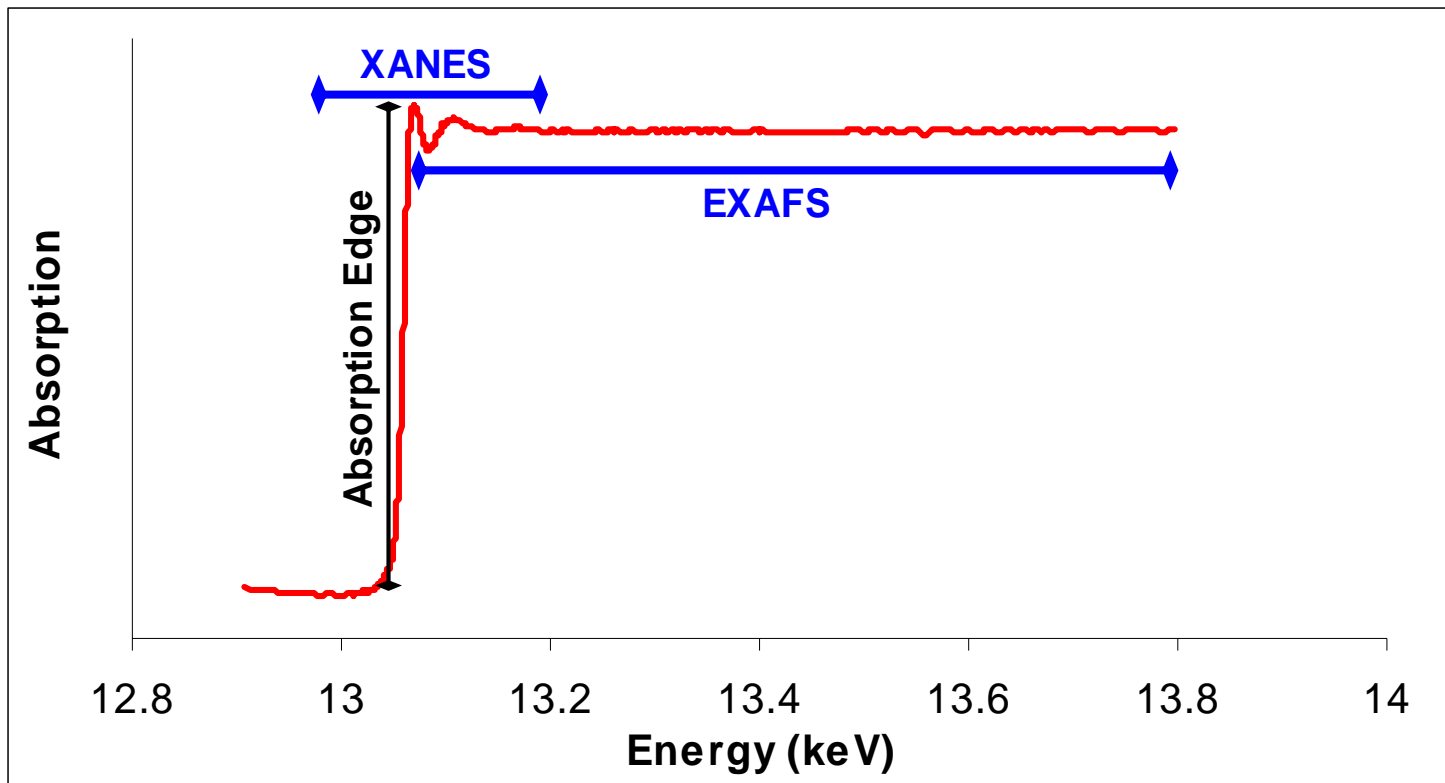


# Factor breakdown

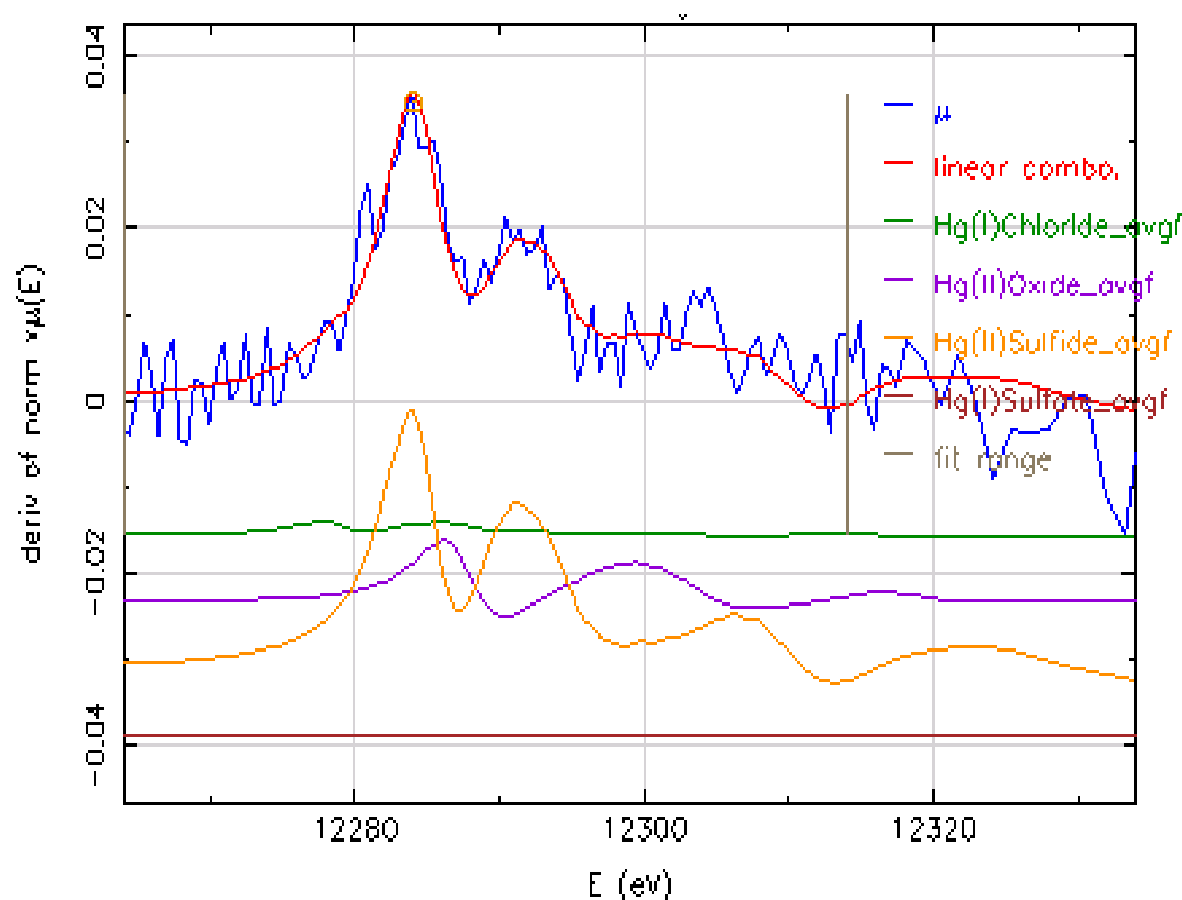
- Factor 1 (41.4%): Carbon, Ca (CEC), P, S, and other heavy metals
- Factor 2 (24.7 %): Al, K
- Factor 3 (7.4 %): Ni
- Factor 4 (6.9%): Cr, Cu, Zn
- Factor 5 (6.5%): Si
- Factor 6 (4.9%): Cd, Zn
- Factor 7 (3.2%): Mg
- Factor 8 (2.9%): S



# X-ray absorption spectroscopy



# Soil Hg speciation (XANES)

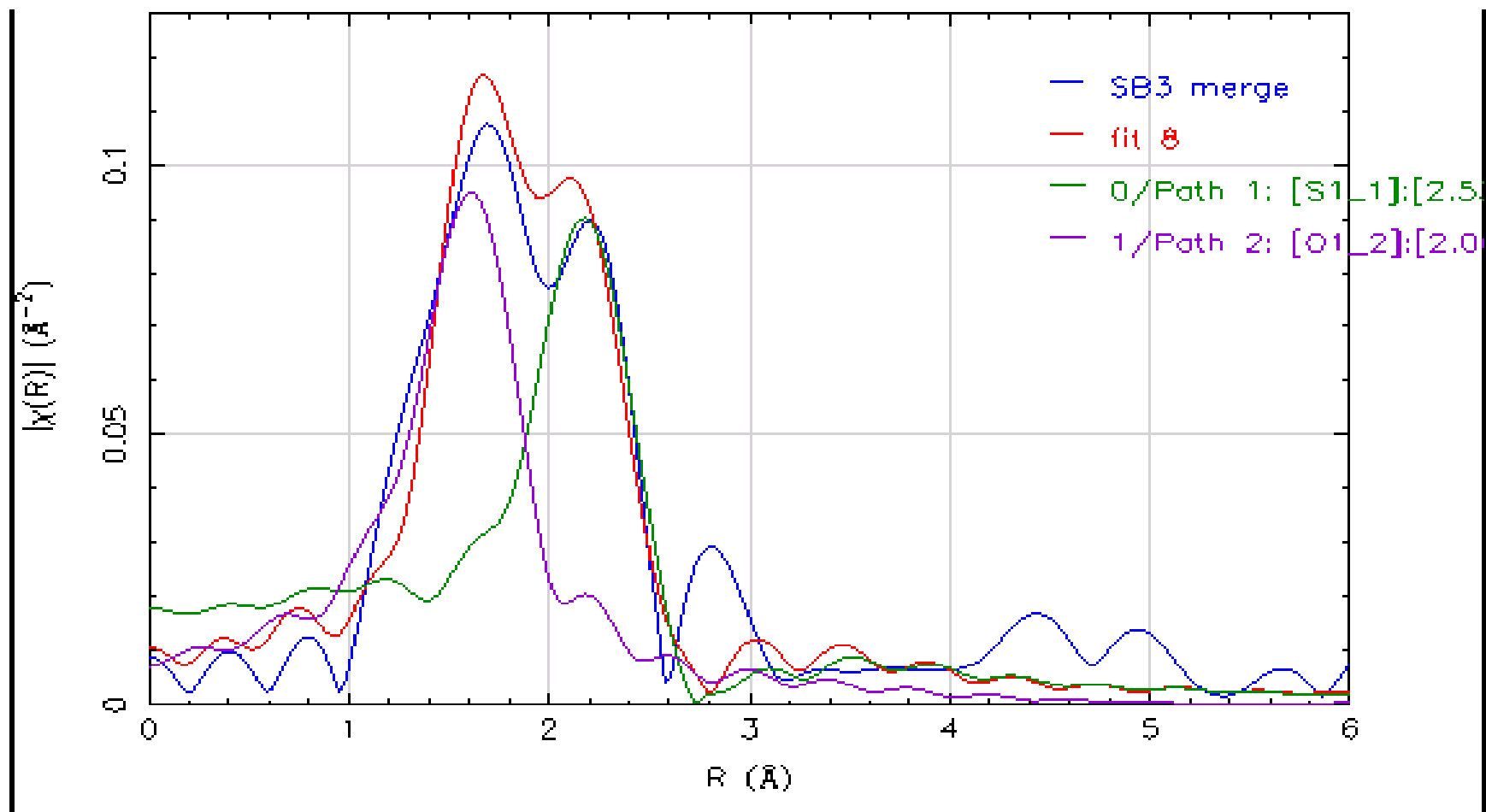


LCF Fits:

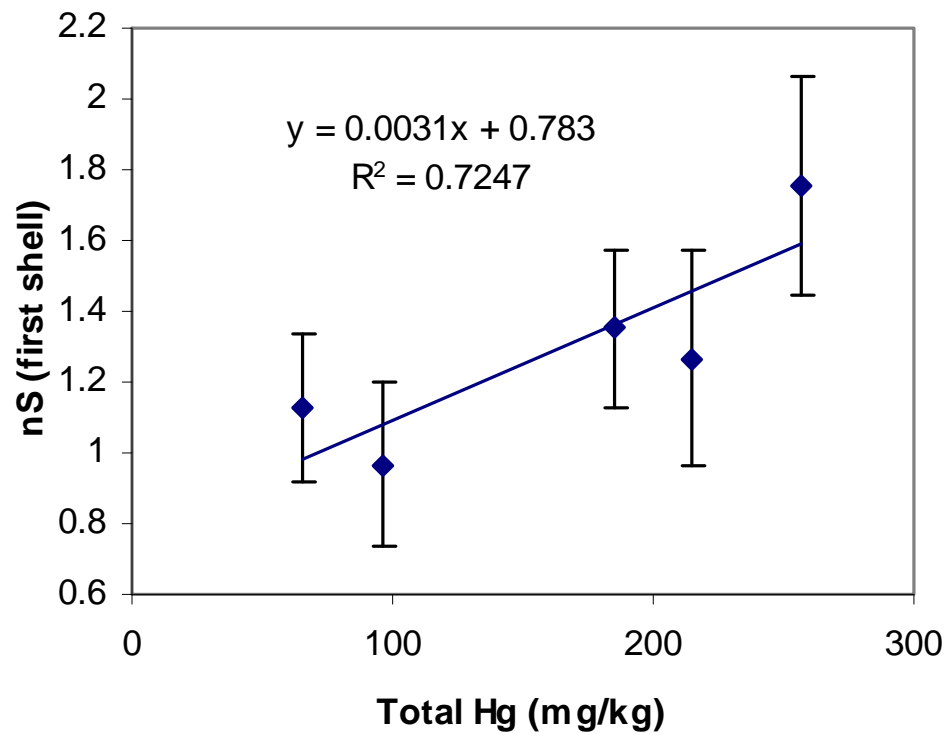
<u>Std</u>	<u>fraction</u>
Hg(I)Cl	0.051
Hg(II)O	0.170
Hg(II)S	0.802

R-factor = 0.082450

# Soil Hg coordination (EXAFS)

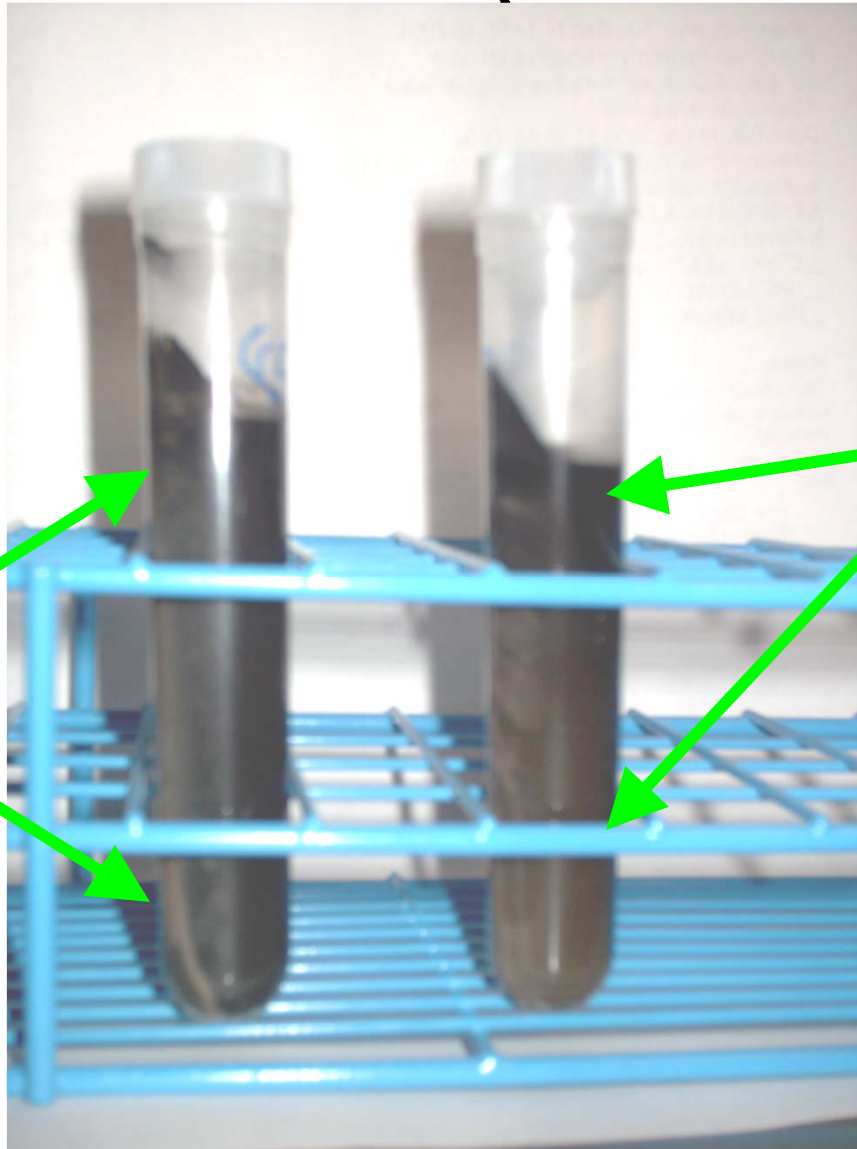


# Organic S coordination





# Surface speciation: Density separation



**Dull-brown  
precipitate  
and streak**

**Crystalline-black  
layer and streak**

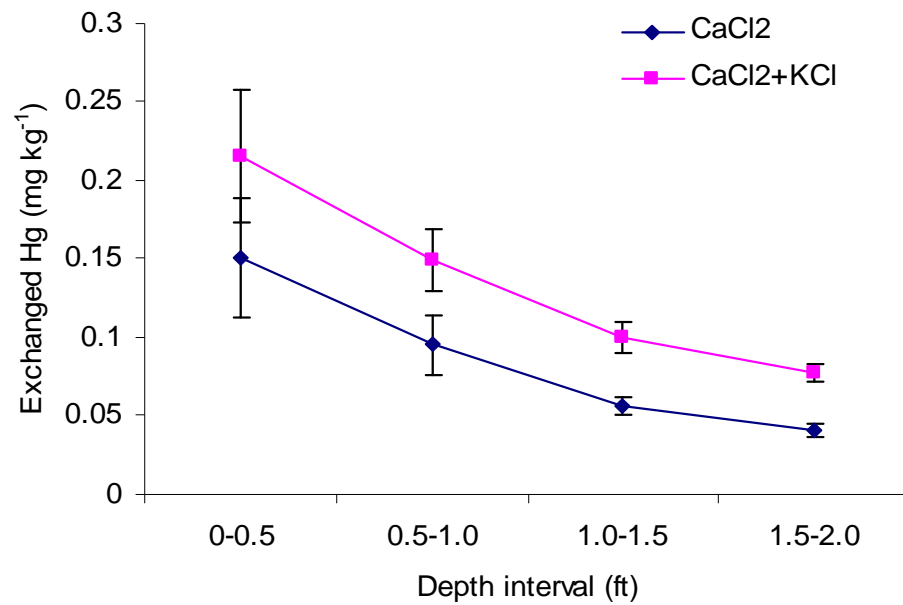
1.5 g cm<sup>-3</sup>  
sodium  
polytungstate  
solution

# Surface speciation: Density separations ( $1.5 \text{ g cm}^{-3}$ )

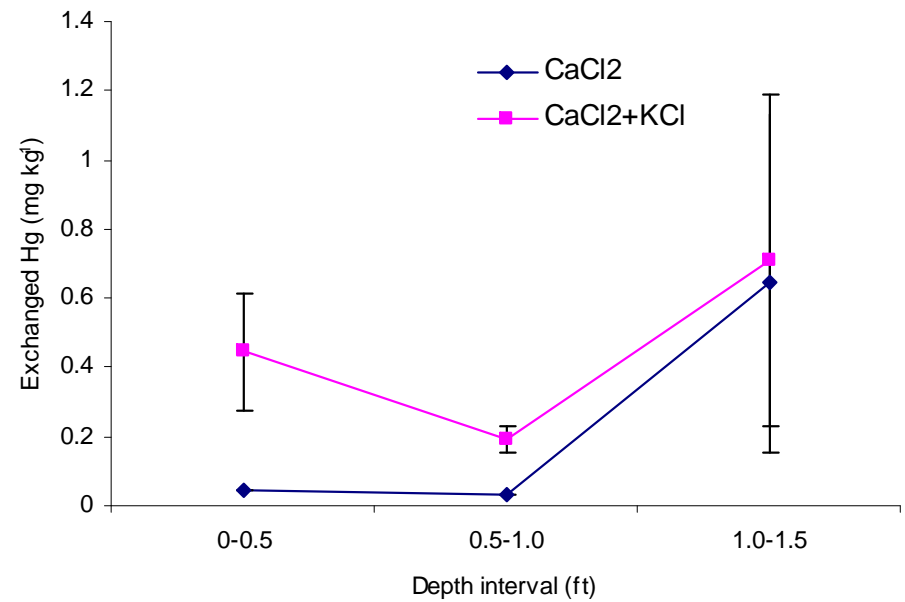
Sample	Depth	Density Fraction	TC	Hg conc.
	Ft	$1.5 \text{ g cm}^{-3}$	%	$\text{mg kg}^{-1}$
SB4	0 – 0.5	low	20.6	533
		high	1.3	110
	0.5-1.0	low	<i>nd</i>	516
		high	1.6	2
	1.0-1.5	low	<i>nd</i>	871
		high	0.6	1

# Exchange reactions

**SB12**



**SB4**



# Exchange chemistry

For the exchange reaction,



the Gapon Exchange coefficient is defined as

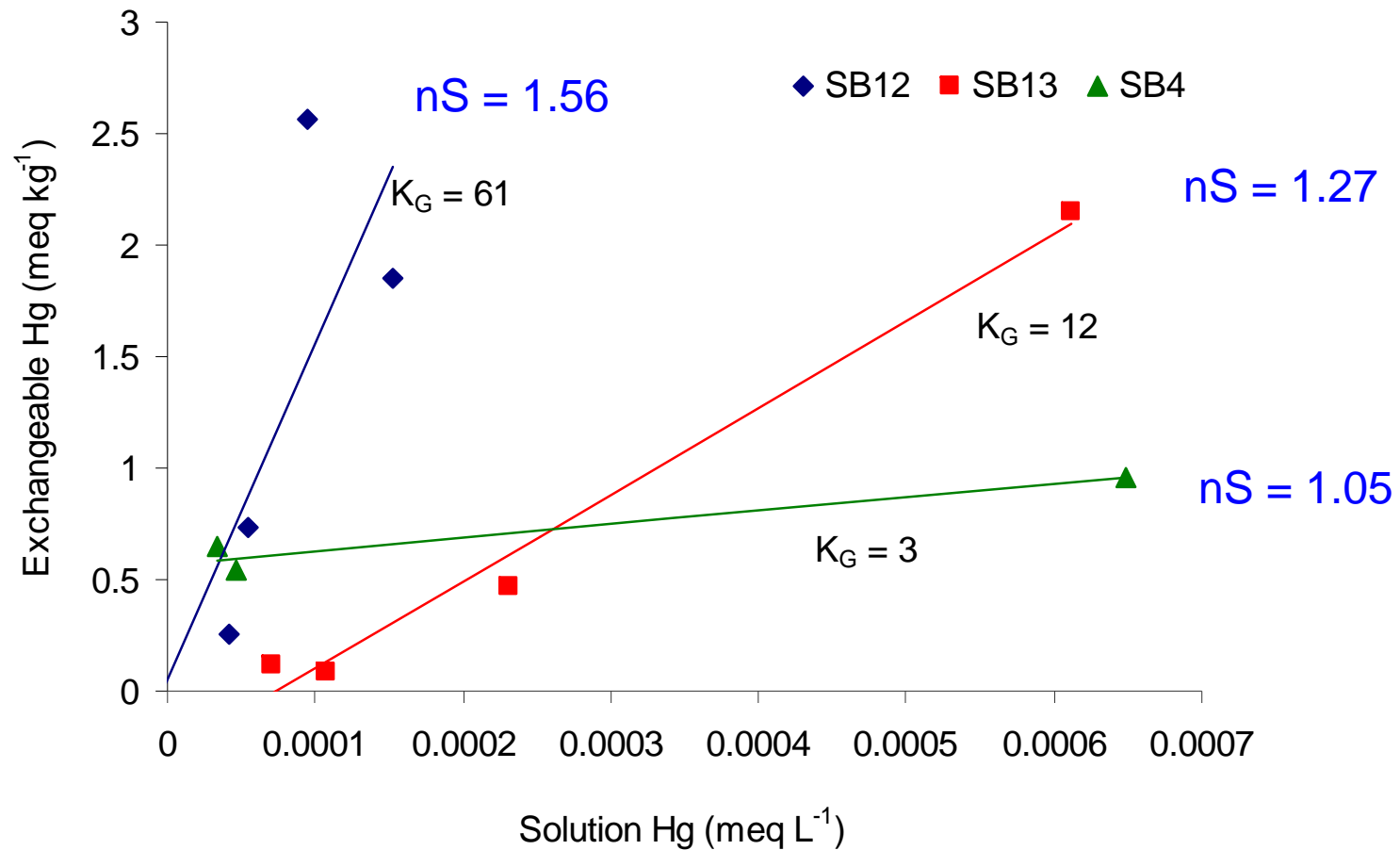
$$K_G = (\text{ExCa}/\text{ExHg}) (\text{Ca}^{2+} / \text{Hg}^{2+})$$

Through some simple assumptions,  
exchangeable Hg can be expressed as,

$$\text{ExHg} = K_G \text{CEC} (\text{Hg}^{2+})$$



# Gapon exchange constants



# Conclusions

## Soil Hg is:

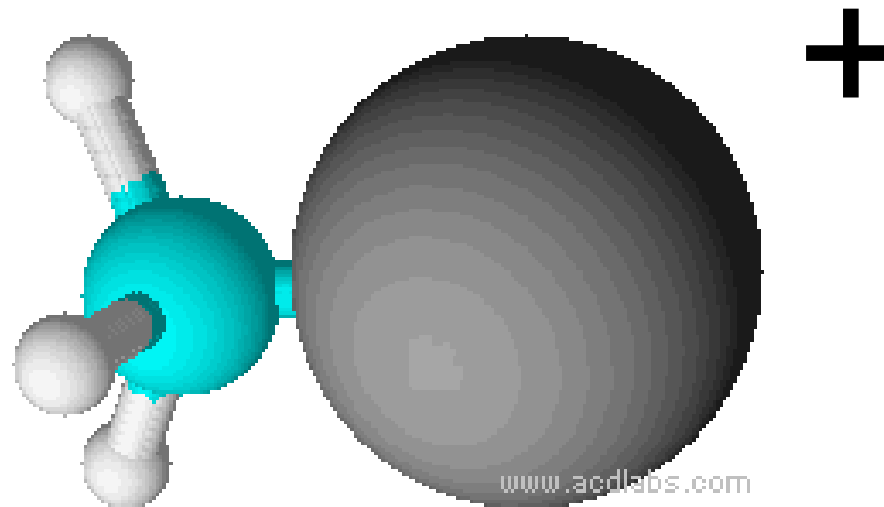
- Present as divalent cation.
  - Readily adsorbed to soil exchange phase.
- Physically associated with the organic carbon fraction.
- Poorly exchangeable due to strong complexation, particularly organic S.
  - Suggests soil Hg is stable under abiotic conditions

# Conclusions

- Low release of Hg during flooding, mostly in the form of tightly adsorbed Hg on dispersed particulates at soil surface.
- Biotic activity spikes with enhanced adsorbed nutrient availability – incidental Hg methylation while C consumed.
- Results suggest soils contain very long-term supply of Hg for methylation for many years to come.

# Questions

- If Hg methylation occurs in soil, what is the fate of MeHg?



Hydrophobic – adsorb to hydrophobic domains

Cationic – adsorb to exchange phase



# FY08 goals

- Investigate in-situ immobilization in soil
  - High affinity for Hg limits bioavailability
  - Non-toxic yet stable
  - Inexpensive & easy to apply